- (21) Application No 8403540
- (22) Date of filing 10 Feb 1984
- (30) Priority data
- (31) 8300736
- (32) 11 Feb 1983
- (33) Sweden (SE)
- (43) Application published 15 Aug 1984
- (51) INT CL³
 C07D 401/00
 A61K 31/435
 C07D 405/14 487/04
 491/04
 (C07D 401/00 217/00 295/
 18)
 (C07D 405/14 213/04 235/
 02 307/04 317/10)
 (C07D 487/04 235/00)
 (C07D 491/04 307/00 317/
- (52) Domestic classification C2C 1173 1174 1175 1354 1416 1418 141X 1426 142X 1472 1485 1492 1530 1532 1535 155X 200 202 211 213 214 215 220 221 225 226 22Y 246 247 250 251 252 253 255 25Y 28X 29X 29Y 305 30Y 311 313 314 31Y 321 323 326 32Y 332 337 339 342 34Y 350 351 352 355 360 361 364 366 368 36Y 371 373 37Y 397 43X 440 461 462 551 574 584 594 601 614 620 623 624 625 628 62X 634 635 644 650 652 655 656 658 65X 662 665 668 671 672 675 676 678 694 698 699 760 776 777 802 80Y AA QL QS RE RM RQ RV WB WC WE WJ ZF ZH U1S 1318 C2C
- (56) Documents cited GB 1525958 GB 1500043 EP 0005129A1 EP 0074341A1
- (58) Field of search C2C
- (71) Applicants
 Aktiebolaget Hassle,
 (Sweden),
 S-431 83 Moindal,
 Sweden
- (72) Inventors
 Arne Elof Brandstrom
 Stig Ake Ingemar
 Carlsson
 Britt Inger Monica
 Kallsson
 Per Lennart Lindberg

74) Agent and/or
Address for Service
J. A. Kemp & Co.,
14 South Square,
Gray's Inn,
London WC1R 5EU

- (54) Novel pharmacologically active compounds
- (57) Novel compounds of the formula:

wherein X is S or SO and R¹, R², R³, R⁴, R⁵, R⁶, R⁷, R⁸ and R¹⁵ are organic residues, pharmaceutical compositions containing such compounds particularly for use in the treatment of gastric disorders.

1

Novel pharmacologically active compounds

The object of the present invention is to provide novel compounds, and therapeutically acceptable salts thereof, which inhibit exogenously or endogenously stimulated gastric acid secretion and provide gastrointestinal cytoprotective effects and thus can
 be used in the prevention and treatment of peptic ulser

The present invention relates to the use of the compounds of the invention or therapeutically acceptable salts thereof, for inhibiting gastric acid

15 secretion as well as providing gastrointestinal cytoprotective effects in mammals and man. In a more general sense, the compounds of the invention may be used for prevention and treatment of gastrointestinal inflammatory diseases in mammals and man,

20 including e.g. gastritis, gastric ulcer, and duodenal ulcer. Furthermore, the compounds may be used for prevention and treatment of other gastrointestinal disorders, where cytoprotective and/or gastric antisecretory effect is desirable e.g. in patients with

25 gastrinomas, in patients with acute upper gastrointestinal bleeding, and in patients with a history of chronic and excessive ethanol consumption. The invention also relates to pharmaceutical compositions containing at least one compound of the

30 invention, or a therapeutically acceptable salt thereof, as active ingredient. In a further aspect, the invention relates to processes for preparation of such new compounds and to novel intermediates in the preparation of the compounds of the invention.

Benzimidazole derivatives intended for inhibiting gastric acid secretion are disclosed in the British patent specifications 1 500 043 and 1 525 958, in the US patent 4 182 766, in the European patent specification 0 005 129, and in the Belgian patent specification 890 024. Benzimidazole derivatives proposed for use in the treatment or prevention of special gastrointestinal inflammatory disease are disclosed in the European patent application with publication no. 0 045 200

45 It has been found that the compounds of the formula

$$\begin{array}{c}
R^{2} \\
\downarrow 5 \\
\downarrow 7 \\
\downarrow$$

wherein

O

$$\uparrow$$

X is —S—or —S—;

 R^{15} is H , CH_3 or C_2H_5 ;

50 R1, R2, R3 and R4, which are the same or different, are

(a) H

(b) halogen

(c) -CN

(d) ---CHO

55 (e) ---CF₃

(h) —CH(OR¹³)₂

(i) $-(Z)_n-A-D$

60 (j) aryl

(k) aryloxy

(I) alkylthio containing 1-6 carbon atoms

(m) -NO₂

(n) alkylsulfinyl containing 1-6 carbon atoms

65 orwherein

(o) adjacent groups R¹, R², R³ and R⁴ together with the adjacent carbon atoms in the benzimidazole ring form a 5-, 6- or 7-membered monocyclic ring or a 9-, 10- or 11-membered bicyclic ring which rings may be saturated or unsaturated and may contain 0-3 hetero atoms selected from N and O, and which rings may be optionally substituted with 1-4 substituents selected from alkyl groups with 1-3 carbon atoms, alkylene radicals containing 4-5 carbon atoms giving spiro
 compounds, or two or four of these substituents together form one or two oxo groups

(—C—), whereby if R¹, R², R³ and R⁴ together with the adjacent carbon atoms in the benzimidazole ring form two rings they may be condensed with each other, in which formulas R¹¹ and R¹², which are the same or different, are

(a) aryi,

0

(b) alkoxy containing 1-4 carbon atoms,

(c) alkoxyalkoxy containing 1-3 carbon atoms in each alkoxy part,

(d) arylalkoxy containing 1-2 carbon atoms in the alkoxy part,

(e) aryloxy,

(f) dialkylamino containing 1-3 carbon atoms in the go alkyl parts, or

(g) pyrrolidino or piperidino, optionally substituted with alkyl containing 1-3 carbon atoms

R¹³ is (a) alkyl containing 1-4 carbon atoms, or (b) alkylene containing 2-3 carbon atoms;

n is 0 or 1;

105

A is (a) alkylene containing 1-6 carbon atoms

(b) cycloalkylene containing 3-6 carbon atoms

(c) alkenylene containing 2-6 carbon atoms

100 (d) cycloalkenylene containing 3-6 carbon atoms,

(e) alkynylene containing 2-6 carbon atoms;
Dis (a) —CN

R9 is (a) alkoxy containing 1-5 carbon atoms, or

```
(b) dialkylamino containing 1-3 carbon atoms in
    the alkyl parts;
      m is 0 or 1;
      ris0or1;
      Yis (a) -- O-
      (b) --NH--
      (c) -NR10-:
      R<sup>10</sup> is (a) H
      (b) alkyl containing 1-3 carbon atoms,
      (c) arylalkyl containing 1-2 carbon atoms in the
    alkyl part, or
      (d) aryl;
      R<sup>5</sup>is(a) Hor
            0
      (b) -
15 wherein
      R<sup>14</sup> is (a) alkyl containing 1-6 carbon atoms,
      (b) arylalkyl containing 1-2 carbon atoms in the
    alkyl part
      (c) arvi
      (d) alkoxy containing 1-4 carbon atoms
      (e) arylalkoxy containing 1-2 carbon atoms in the
    aikyi part
      (f) aryloxy
      (g) amino
      (h) mono- or dialkylamino containing 1-4 carbon
    atoms in the alkyl part(s)
      (i) arylalkylamino containing 1-2 carbon atoms in
    the alkyl part
      (j) arylamino;
      R<sup>6</sup> and R<sup>8</sup>, which are the same or different, are
30
      (a) Hor
      (b) alkyl containing 1-5 carbon atoms;
       R7 is (a) H
       (b) alkyl containing 1-8 carbon atoms
      (c) alkoxy containing 1-8 carbon atoms
35
       (d) alkenyloxy containing 2-5 carbon atoms
       (e) alkynyloxy containing 2-5 carbon atoms
       (f) alkoxyalkoxy containing 1-2 carbon atoms in
    each alkoxy group
      (g) dialkylaminoalkoxy containing 1-2 carbon
    atoms in the alkyl substituents on the amino nitrogen
    and 1-4 carbon atoms in the alkoxy group
       (h) oxacycloalkyl containing one oxygen atom and
    3-7 carbon atoms
       (i) oxacycloalkoxy containing two oxygen atoms
    and 4-7 carbon atoms
       (j) oxacycloalkylalkyl containing one oxygen atom
    and 4-7 carbon atoms
       (k) oxacycloalkylalkoxy containing two oxygen
 50 atoms and 4-6 carbon atoms, or
       (I) R^6 and R^7, or R^7 and R^8 together with the
     adjacent carbon atoms in the pyridine ring form a ring
     wherein the part constituted by R<sup>6</sup> and R<sup>7</sup>, or R<sup>7</sup> and
     R<sup>8</sup>, is
         -CH=CH-CH=CH
         –O---(CH₂)<sub>p</sub>---
         -CH₂(CH₂)<sub>P</sub>-
       --O--CH=CH--
        ---NH---CH=CH---
        -N-CH=CH-
 60
```

wherein p is 2, 3 or 4 and the 0 and N atoms always

CH₃

are attached to position 4 in the pyridine ring; and physiologically acceptable salts of the com-65 pounds I wherein X is S;

with the provisos that

(a) not more than one of R^6 , R^7 and R^8 is hydrogen,

(b) when X is SO, R⁵ is H and R⁶, R⁷ and R⁸ are selected only from hydrogen, methyl, methoxy, 70 ethoxy, methoxyethoxy and ethoxyethoxy and at the same time more than one of R1, R2, R3 and R4 are hydrogen, then R1, R2, R3 and R4 cannot be selected

only from alkyl groups, halogen, alkoxycarbonyl, alkoxy or alkanoyl,

(c) when X is S, R⁵ is H, alkanoyl or alkoxycarbonyl, and R⁶, R⁷ and R⁸ are selected only from hydrogen, methyl, ethyl, methoxy, ethoxy, methoxyethoxy and ethoxyethoxy and at the same time more than one of R^1 , R^2 , R^3 and R^4 are hydrogen, then R^1 , R^2 , R^3 and R^4 . 80 cannot be selected only from alkyl groups, halogen, alkoxycarbonyl, alkoxy, alkanoyl, trifluormethyl, or NO₂,

(d) when X is SO, one of R⁶, R⁷ and R⁸ is H and the other two of R⁶, R⁷ and R⁸ are alkyl, and at the same 85 time more than one of R¹, R², R³ and R⁴ are hydrogen, then those radicals R1, R2, R3 and R4 which are not H cannot be selected only from alkyl, halogen, cyano,

(e) when R3, R4, R5 and R15 are H and simultaneously R^6 and R^8 are H or CH_3 and R^7 is OCH_3 , then R^1 is not CF₃ when R² is H, and R² is not CF₃ when R¹ is H, are effective as gastrointestinal cytoprotectives and as inhibitors of gastric acid secretion in mammals 95 and man as stated above.

Illustrative examples of the various radicals in the formula lare as follows. These illustrative examples will be applicable to different radicals depending on the number of carbon atoms prescribed for each 100 radical. It will be understood that the expressions "alkyl" and "alkoxy" include straight, branched and cyclic structures.

Aixy1:
$$CH_3$$
, C_2H_5 , $n-C_3H_7$, $i-C_3H_7$, $n-C_4H_9$, $sec.-C_4H_9$, $iso.-C_4H_9$, $tert.-C_4H_9$, $n-C_5H_{11}$, $n-C_6H_{13}$, $-CH_2$,

Alkoxyalkoxy: -och₂och₃, -och₂ch₂och₂ch₃, -och₂ch₂och₂ch₃,

Alkanyloxy: -0-CH-CH₂ , -0-CH-CH-CH₃ , -0-CH-CH-C₂H₅, -0-CH₂-CH-CH-CH₂CH₃

. It lustrative examples of the radical -CH: $3e^{\frac{1}{3}})_2$ are:

Illustrative examples of the ring structures involving π^2 , R^3 or R^4 are

where w is

-сн₂сн₂сн₂--ch2ch2ch2ch2--CH2-C(CH3)2-CH2--(ch₂)₅--CH=CH-CH=CH--сн-сн₂сн₂--ch-ch-ch-ch--ch3 ch3 -ch-ch-ch-ch--ch-ch-ch-ch--ch3 ch3 -ch3 ch3 -ch3 ch3 -(сн₂)₂-мн-

> -осн₂сн₂о--о-с(сн₃)₂-о--0(CH₂)30-

The radical $-(Z)_{n}$ - A - O comprises the following radicals. The expression (alkyl 1-3c) etc. means alkyl groups containing 1, 2 or 3 carbon atoms.

```
A - CN
   )
A - C - 9 -(alkyl 1-5c)
                {alkyl 1-3c}
                (alkyl 1-3c)
   A - (alkyl 1-3c)
   A - (alkyl 1-2c)-aryl
   A - aryl
  A - 0 - H
  A - 0 -(alkyl 1-3c)
  A - 0 -(alkyi 1-2c)-aryl
  A = 0 - aryl
  A - AH - H
 A - NH - (alkyl 1-3c)
  A - NH -(alkyl 1-2c)-aryl
 A - NH - aryl
RIG
A - N - H
 R 10
A - N -(alkyl 1-3c)
 A - N -Calkyl 1-2c)-aryl
 A - N - aryl
 A - O - C - H
 0
A- 0 - C - (alkyl 1-3c)
 A- 0 - C - (alkyl l-2c)-aryl
 A- 0 - C - aryl
 A- NH - C - H
 q
A- NH - C -(alkyl 1-3c)
 A- NH - C -(alkyl 1-2c)-aryl
 A- NH - eryl
 R10. C
1 IE
A- N - C - H
 R<sup>1G</sup> O
A - N - C - (alkyl 1-3c)
 R10 0
A - N - C -(alkyl 1-2c)-aryl
 R10 0
A - N - C -aryl
 -G -A - CN
 -0 -A - C-O-(alkyl 1-Sc)
-0 -A - C - N (alkyl 1-3c)
                 (alkyl 1-3c)
-Q-A - H
-0-A -(alkyl 1-3c)
-0-A-(alkyl 1-2c)-aryl
-Q-A-aryl
-0 - A - O - H
-0 - A - 0 -(alkyl 1-3c)
-0 - A - 0 -(alkyl 1-2c)-aryl
-0 - A - 0 - aryl
-0 - A - NH - H
```

```
-0 - A - NH -(alkyl 1-3c)
   -0 - A - NH -(alkyl i-2c)-aryl
-0 - A - NH - aryl
              a10
   -0 - A - N - H
   R<sup>10</sup>
-0 - A - N -(alkyl 1-3c)
             Ŗ<sup>10</sup>
   -0 - A - N -(alkyl 1-2c)-aryl
             R<sup>10</sup>
  -0 - A - N - aryl
  -0 - A - 0 - c - (alkyl 1-3c)
  -0 - A - 0- C - (alkyl 1-2c)-aryl
  0
-0 - A - 0- C - aryl
-0 - A - NH - C - H
-0 - A - NH - C - (alkyl 1-3c)
-0 - A - Nr - C -(alkyl 1-2c)-aryl
-0 - A - NH - aryl
          Ŗ<sup>10</sup>
R^{10} 0 - 0 - A - N - C - (alkyl 1-3c)
R 10 0
-0 - A - N - C -(alkyl 1-2c)-aryl
R<sup>10</sup> 0
-0 - A - N - C -aryl
-C- A -CN
0 0
-C- A -C - O-(alkyl 1-5c)
                 /(alkyl 1-3c)
                  (alkyl 1-3c)
 -C -A -H
 0
|-C -A -(alkyl 1-3c)
 0
|-C -A -(alkyl 1-2c)-aryl
 0
-C -A- aryl
```

```
-- Э
-- С -А -О -н
            0
||
|-C -A -O-(alkyl 1-3c)
            0
-C -A -O -aryl
            0 .
-C -A -NH -H
            0
-C -A -NH -(alkyl 1-3c)
           0
-C -A -NH -(alkyl 1-2c) -aryl
           0
||
|C -A -NH -aryl
          O R10
-C -A -N -(alkyl 1-3c)
-C -A -N -(alkyl 1-2c)-aryl
-C -A -N -aryl
           0
-С-А-О-С-Н
           0 0
||
-C-A-J-C-(alkyl 1-3C)
            0 0
|| -C-A-O-C-(alkyl 1-2C)-aryl
            0 0
|| ||
-C-A-O-C-aryl
            0 0
-C-A-NH-C-H
            -0 0
| -C-A-NH-C-(alkyl 1-3C)
            O R<sup>10</sup> O
|| | | | |
|-C-A-N - C-(alkyl 1-3C)
             0 R<sup>10</sup> 0
| | | | |
|-C-A-N - C-(alkyl 1-2C)-aryl
            o R<sup>10</sup> 0
-C-A-N - C-aryl
The radical \overset{Q}{\sim}-R comprises the following radicals.
```

```
g
-C-aryl
                  0
-C-0-(alkyl 1-40)
                   0
-C-O-(alkyl 1-3 c)-O-(alkyl 1-3c)
                   0
-C-O-(alky1 1-2c)-ary1
                   0
-C-N (alkyl 1-3c)
(alkyl 1-3c)
        -C-f( (optionally substituted with alkyi)
        -C-N (optionally substituted with alkyl)
The radical -0-\dot{c}-R^{12} comprises the following radicals.
        Q
-O-C-aryl
        0
-0-C-0-(alkyl 1-4c)
        0
-0-C-0-(alkyl 1-3c)-0-(alkyl 1-3c)
         -0-C-(alkyl 1-2c)-aryl
         -0-C-0-aryl
         0 (alkyl 1-3c)
-0-C- (alkyl 1-3c)
         \begin{array}{c} 0 \\ -0-C-N \end{array} \hspace{0.5cm} \text{ (optionally substituted with alkyl)}
                     (optionally substituted with alkyl)
The radical -C-R<sup>14</sup> comprises the following radicals:
        -C-(alkyl 1-6c)
        0
-C-(alkyl 1-2c)-aryl
         0
-C-O-(alkyl 1-4c)
         0
-C-0-(alkyl l-2c)-aryl
         -ċ-0-aryl
         о
-с-пн<sub>2</sub>
         -C-NH(alkyl 1-4cl
          0
(alkyl 1-4c)
(alkyl 1-4c)
         0
n (alkyl 1-2c)
-C-f
               aryl
         0
-C-NH(aryl)
```

Further illustrative examples of the radicals in the formula I are:

alkylsulfinyl:

SOCH₃, SOC₂H₅, SOCH₂CH₂CH₃, SO-i-C₃H₇, SO-n-C₄H₉, SO-n-C₅H₁₁

50-n-C4Hg, 50-n-C5H

oxacycloalkyl:

 \sim

oxacycloalkoxy:

<u>_____</u>

oxacycloalkyl-alkyl:

CH2-

oxacycloalkyl-alkoxy:

OCH2-0-

The compounds of the invention that are sulfoxides (X=SO) have an asymmetric centre in the sulfur atom, i.e. these compounds exist as two optical isomers (enantiomers), or if they also contain one or 5 more asymmetric carbon atoms the compounds have two or more diastereomeric forms, each existing in two enantiomeric forms. Such asymmetric carbon atoms may be the carbon atom on which R¹⁵ is attached (when R¹⁵ is other than H) or a carbon atom 10 in some of the substituents.

Both the pure enantiomers, racemic mixtures (50% of each enantiomer) and unequal mixture of the two are within the scope of the present invention. It should be understood that all the diastereomeric forms possible (notes appears as a second in the companies).

15 forms possible (pure enantioners or racemic mixtures) are within the scope of the invention.

The compounds of the invention that are sulfides (X=S) may be asymmetric due to one or more asymmetric carbon atoms, as described above. The 20 different diasetereomeric forms possible as well as the pure enantiomers and racemic mixtures are within the scope of the invention.

It should be noted that for all the compounds of the invention wherein R⁵ is H the substituents R¹ and R⁴
25 as well as R² and R³ are considered to be equivalent. This is due to the tautomerism in the imidazole part of the benzimidazole nucleus causing an equilibrium between the two possible NH-forms. This is illustrated by the following example:

- 30 I Preferred groups of the radicals R¹, R², R³ and R⁴ are:
 - 1. H
 - 2. halogens F, CI, Br and the groups CN, CHO, CO(aryl), COO(alkyl), CF₃, SCH₃, SOCH₃ and NO₂
- 35 3. the groups alkylene-D, O-alkylene-D and CO-alkylene-D wherein D is CN, COO(alkyl), COR¹⁰, OR¹⁰ and R¹⁰
 - 4. aryl and aryloxy

- 40 6. —CH₂CH₂CH₂—, —CH₂CH₂CH₂CH₂—and —CH=CH—CH=CH—
 - 7. —CH=CH—CH=C—(CH₂)₂₋₃—
 - 8. saturated heterocyclic ring structures having 2
- 45 oxygen atoms.
 - 9. unsaturated 6-membered heterocyclic ring structures having one nitrogen atom
 - II Further preferred groups of the radicals R^1 , R^2 , R^3 and R^4 are:
- 50 1. H
 - halogens Cl and Br and the groups CO(phenyl), COOCH₃, CF₃, SCH₃ and SOCH₃
 - 3. the groups alkyl, alkoxyalkyl, aryloxyalkyl, arylaklyl, aryl
- 4. the groups alkoxy, alkoxyalkoxy, aryloxyalkoxy, aryloxy
 - 5. the group alkanoyl
 - 6. —CH₂CH₂CH₂—,—CH₂CH₂CH₂CH₂—and —CH=CH—CH=CH—
- 60 | 7. --CH=CH--CH=C--(CH₂)₂₋₃--
 - 8. saturated heterocyclic ring structures having 2 oxygen atoms in 4,5-, 5,6- or 6,7-"catechol positions", e.g. (5,6-position shown)

- 65 III Still further preferred groups of the radicals R¹, R², R³ and R⁴ are:
 - 1. H
 - 2. Br and the groups COOCH3 and CF3
- 3. the groups CH₃, C₂H₅, CH(CH₃)₂, CH₃OCH₂CH₂—, 70 phenyl
- the groups CH₃O, CH₃(CH₂)₆O—, CH₃OCH₂CH₂O—, (phenyl)- OCH₂CH₂CH₂O—, (phenyl)CH₂CH₂O—, (phenyl)O—
 - 5. the groups CH₃CO—, C₂H₅CO—
- 75 6. —CH₂CH₂CH₂—,—CH₂CH₂CH₂CH₂—
 - 7. —OCH₂O—, -0 0- in the 5,6-"catechol position"
 - IV Particularly preferred groups of the radicals R^1 , R^2 , R^3 and R^4 are:
 - H, COOCH₃, CF₃, CH₃, C₂H₅, CH(CH₃)₃, CH₃O,
- 80 —CH₂CH₂CH₂—, —CH₂CH₂CH₂CH₂— and —OCH₂O— V In a preferred embodiment, at least three of the radicals R¹, R², R³ and R⁴ are other than hydrogen, or

they form at least one ring.

VI In another preferred embodiment the radicals R¹ and R² form a ring structure

VII In another preferred embodiment the radicals R² and R³ form a ring structure.

VIII In a preferred embodiment at least three of the radicals R^1 , R^2 , R^3 and R^4 are other than hydrogen. IX In a preferred embodiment the radicals R^1 , R^2 , R^3 and R^4 are selected from H, halogen, CF_3 , alkyl and alkoxy groups.

X In a preferred embodiment the radicals R^1 , R^2 , R^3 and R^4 are selected from H, alkyl and alkoxy groups. XI In a preferred embodiment the radicals R^1 , R^2 , R^3 and R^4 are selected from H and alkyl groups.

15 XII The preferred groups of X is S.
XIII The preferred group of X is SO.

XIV The preferred group of R¹⁵ is H.

XV Preferred groups of the radical R⁵ are H, arylcarbonyl, alkoxycarbonyl, arylalkoxycarbonyl, di-

alkylaminocarbonyl and arylaminocarbonyl.
XVI Further preferred groups of the radical R⁵ are H, phenylcarbonyl, methoxycarbonyl, tert-butoxycarbonyl, benzyloxycarbonyl, dimethylaminocarbonyl and phenylaminocarbonyl.

XVII Particularly preferred of the radical R⁵ is H.
 XVII Preferred groups of the radicals R⁶ and R⁸ are:
 H, CH₃, C₂H25, C₃H₇ and CH(CH₃)₂

2. ring structures connecting position 4 in the pyridine ring.

30 XIX Particularly preferred groups of the radicals R⁶ and R⁸ are H, CH₃, C₂H₅ and ring structures also connecting position 4 in the pyridine ring XX Preferred groups of the radical R⁷ are:

H, CH₃, C₂H₅

35 2. OCH₃, OC₂H₅, OCH₂CH₂CH₃, O(CH₂)₃CH₃, OCH₂

OCH₂CH=CH₂, OCH₂C≡CH

5. OCH₂CH₂N(CH₃)₂

6. —CH=CH—CH=CH-bound to positions 3 and 4,

-CH=CH-CH=CH-bound to positions 4 and 5,

---CH₂CH₂CH₂-bound to positions 3 and 4,

-CH₂CH₂CH₂-bound to positions 4 and 5,

-CH₂CH₂CH₂CH₂-bound to positions 3 and 4,

-CH2CH2CH2CH2-bound to positions 4 and 5,

—OCH₂CH₂-bound to positions 3 and 4,

-OCH₂CH₂-bound to positions 4 and 5,

-OCH₂CH₂CH₂-bound to positions 3 and 4,

-OCH₂CH₂CH₂-bound to positions 4 and 5,

XXI Further preferred groups of the radical R7 are:

50 1. CH₃

4 and 5.

40

45

2. OCH₃, OC₂H₅, OCH₂CH₂CH(CH₃)₂

3. OCH₂CH=CH₂

4. OCH₂CH₂OCH₃, OCH₂,

5. —CH₂CH₂CH₂-bound to positions 3 and 4,

55 —CH₂CH₂CH₂-bound to positions 4 and 5,

—CH₂CH₂CH₂-bound to positions 3 and 4,

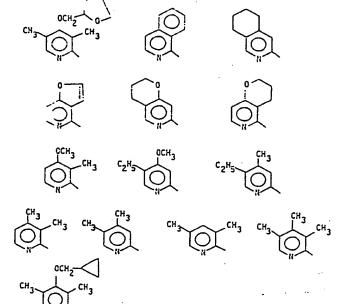
—CH₂CH₂CH₂-bound to positions 4 and 5,

—OCH₂CH₂-bound to positions 3 and 4, —OCH₂CH₂-bound to positions 4 and 5, —OCH₂CH₂-bound to positions 3 and 4, —OCH₂CH₂-bound to positions 3 and 4, —OCH₂CH₂-bound to positions

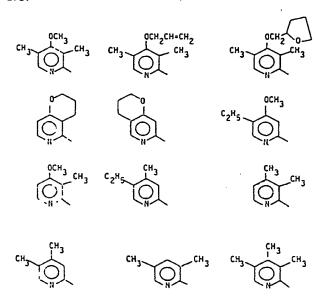
XXII Particularly preferred groups of the radical R^7 are CH_3 , OCH_3 , $OCH_2CH_2CH(CH_3)_2$, $OCH_2 \longrightarrow 0$.

—OCH₂CH₂CH₂-bound to positions 3 and 4 or to positions 4 and 5.

XXIII Preferred pyridyl substitution patterns are:



XXIV Further preferred pyridyl substitution patterns are:



XXV Still further preferred pyridyl substitution patterns are:

XXVI Particularly preferred pyridyl substitution patterns are:

5 XXVII In a preferred embodiment two of the radicals R⁶, R⁷ and R⁸ form one ring structure and the third radical of R⁶, R⁷ and R⁸ is H.

XXVIII In a preferred embodiment R^{15} and R^5 are H, at least three times of the radicals R^1 , R^2 , R^3 and R^4 are 10 other than H, R^6 and R^8 are H or CH₃ and R^6 is CH₃,

_o7

XXIX In a preferred embodiment R¹⁵ and R⁵ are H, the radicals R¹, R², R³ and R⁴ form at least one ring structure, R⁶ and R⁸ are H or CH₃ and R⁷ is CH³, OCH³ or OCH₂CH=CH₂.

XXX Preferred compounds are those of the formula

wherein R^2 is alkyl or alkoxy, preferably CH_3 , C_2H_5 , $CH(CH_3)_2$ and OCH_3 , and X is S or SO.

Further illustrative examples of the radicals in the 20 formula I are given in the examples and lists of specific compounds given elsewhere in this specification.

Illustrative examples of compounds included in the scope of the invention are given in the following

Table 1.

Table 1 Illustrative examples of compounds included in the scope of the invention.

:	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸
i	н	CH3	CH3	CH ³	. СН3	н	CH3	осн2сн=сн5	СН
0	н	CH ₃	CH ³	CH ³	CH3	н	CH3	осн ₂ сн-сн ₂	, CH ₃
	H	CH3	CH3	CH3	CH3	н	CH3	осн ₃	. сн
0	н	CH3	CH ³	СН3	CH ₃	н	CH ₃	осн ₃	• сн _з
	н	CH3	CH ₃	CH ₃	H	н	СН3	OCH2CH=CH2	СН ³
0	H	CH3	CH3	CH3	H	ĸ	CH3	осн ₂ сн-сн ₂	. Сн _з
	H	CH3	CH ₃	CH3	H	н	CH ₃	OCH.	. сн
0	н	CH ³	CH3	сн3	н	н	CH3	осна	CH ₃
	H	CH3	сн ₃	н	CH ₃	н	CH3	CCH ₂ CH=CH ₂	CH ₃
0	н	CH ₃	cH ³	н	CH3	н	CH ₃	OCH ₂ CH=CH ₂	CH3
	н	CH ₃	CH3	н	CH ₃	н	CH ₃	OCH ²	CH3
)	н	CH ₃	сн ₃	н .	CH3	н	СН3	осна	CH ₃
	н	CH3	CH ₃	н	H	н	CH3	осн ₂ сн-сн ₂	CH3

-- cont.

x _	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ^S	R ⁶	R ⁷	88
10	н	CH ³	сн ₃	н	н	н	сн3	OCH ₂ CH=CH ₂	CH3
	H	н	сн3	CH3 ·	H	н	CH3	осн ₂ сн∗сн ₂	СНЗ
0	н	н	снз	CH3	н	H	снз	осн ₂ сн•сн ₂	€н3
	H	CH ₃	н ,	н	CH ₃	н	CH3	осн ² сн•сн ⁵	CH3
0	H	CH3	н	н	CH3	н	CH3	осн ² сн∍сн ⁵	CH3
	H	CH3	н	н	н	н	CH3	осн ₂ сн•сн ₂	CH ³
0	H	CH3	н	н	H	H	CH ³	OCH ² CH*CH ²	СНЗ
	H	н	сн3	н	н	Н	CH3	och ^s ch-ch ^s	CH ₃
)	H	н	сн3	н	н	н	CH3	och ₂ ch=ch ₂	CH3
	н	H	ocH3	н	н	н	CH3	осн ₂ сн=сн ₂	CH3
0	H	H	оснэ	H	н	н	CH3	och ₂ ch=ch ₂	сн3
	H	H	ссн ₃	н	н	н	CH3	och ₂ c≠ch	CH3
0	H	H	OCH3	н	н	н	CH ₃	GCH ⁵ C≡CH	CH3
0	H	H	осн3	н	н	H	CH3	о(сн ⁵) ³ сн*сн ⁵	CH3
0	H	H	ссн3	н	н	Я	CH3	о(сн ⁵) ³ сн ³	CH3
	H	ж .	осн ₃	н	н	H	CH ₃	осн(сн ₃) ²	Сн3
)	H	н	осн3	н	н	н	CH3	осн(сн ₃) ₂	CH3
	н	н	осн3	н	н	н	CH3	ос(сн ₃)3	CH3
)	H	н	осн ₃	н	н	н	CH3	OC(CH ³) ³	CH3
									con

1	R15	R ¹	R ²	R ³	R ⁴	R ⁵	. R ₆	R ⁷	R ⁸
,	н	н	осн3	н	н	н	CH ³	°<> ≥.5	CH3
0	н	н .	осн3	н	н	н	CH ₃	o-🔷	снз
	н	н	CCH3	н	н	н	CH ₃	ссн ₂ —<	CH3
0	н	н	осн3	н	н	H	CH ³	OCH2	CH3
	н	н	OCH3	н	н	H	CH3	OCH ₂ -	CH3
9	н	н	осн3	н	н	н	CH ³	осн ₂	CH ₃
	н	н	осн ₃	н	н	н	CH3	0(CH ₂)2H(CH ₃)2	CH3
	н	н	осн3	н	н	H	CH3	о(сн ₂)2 ^н Дн(сн ³) ² с1 _©	CH ₃
)	н	н	OCH3	н	н	н	CH3	о(сн ²) ² и(сн ³) ²	CH3
	н	н	осн3	н	н	н	CH3	осн ₂ сн ₂ сн(сн ₃) ₂	CH3
0	н	н	осн3	н	н	н	CH ₃	OCH2CH2CH(CH3)2	CH3
0	н	н	GCH ₃	н	н	н	н	осн ₃	C2H5
	н	н	осн3	H	н	н	н	O(CH ₂)3CH3	C2H5
٥	н	н	осн ₃	н	н	н	H	о(сн ₂)3сн3	C ₂ H ₅
0	н	н	OCH ₃	н	н	н	CH3	осн ₂ сн ₂ сн ₂ сн(сн ₃) ₂	CH ₃
0	н	CH ₃	осн ₃	• сн _а	н	я	н	C2H5	7 CH ₃
0	н	H	осн _з	н	н	н	сн ₃	OCH2CH2CH2-	CH3
0	н	CHJ	осн3	сн _З	н	н	н	сн(сн ₃) ₂	CH3

x	R ¹⁵	R ¹	R ²	R ³	x ₁	R ⁵	R ⁵ R ⁷	R ⁸
s	н	н •	есн3	н	н	н	н -(сн ⁵)*	•
so	н	н .	och 3	H	н	н	H -(CH2)4	
;	H	н	оснэ	H	н	H	-(CH ₂)4-	н
0	н	н	осн ³	н	н	н	-(CH ₂)4-	н
	н	H	OCH3	H	н	н	н -0-(сн ₂)3-
3	H	н	оснэ	н	н	н	H +0-(CH ₂	
	н	н	осн3	н	н	н	-(CH ₂) ₂ -0-	н
3	н	н	осн3	н	н	H	-(CH ₂) ₂ -0-	H
	н	н	OCH3	н	н	н	H +CH=CH-CH=	CH-
1	н	н	осн3	н	н	H	H -CH-CH-CH=	CX+
	н	н	осн ₃	н	H	я	-CH-CH-CH-CH-	н
	н	н	осн3	н	н	R	-CH=CH-CH=CH-	Н
	н	н	CH(0)	н	н	н	сн ₃ осн ₃	сн
)	н	н	CH 0	н	н	. н	сн ₃ осн ₃	СН
	H	н	сн(осн ₃) ₂	н	н	н	сн ₃ осн ₃	сн
								con c.

	R ¹⁵	R ³	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸
0	н	н	сн(осн ₃)2	н	н	н	CH3	осн3	СНЗ
	н	н	СНО	Н	н	H	CH ₃	осн3	сн3
0	н	ж	СНО	н	H	H	CH3	осн ₃	CH3
	H	H	CH=CH-COOC2HS	н	н	н	CH3	OCH3	CH3
0	H	н	сн=сн-соос ₂ н ₅	н	H	н	CH3	осн3	CH3
	H	н	CH2CH2COOC2H5	н	н	н	CH3	осн ₃	СНЗ
0	H.	н	CH2CH2COOC2H5	н	H	н	сн3	OCH3	CH3
	К	н	CHSCHScon(CH3)S	H	H	н	CH3	OCH3	CH ³
0	н	н	CH2CH2CON(CH3)2	н	н	н	CH ³	осн3	CH ³
	н	н	CH=CH-CN	н	н	н	CH ³	OCH3	CH3
0	н	н	CH=CH-CH	н	н	н	CH3	осн3	сн
	н	н	CH ² CH ² CH	H	н	н	CH3	осн3	CH3
0	н	н	CH2CH2CN	н	н	н	CH3	осн3	CH3
	н	н	сн ² сн ² сн ² ан	н	H	н	сн ₃	осн ₃	CH3
0	н	н	сн ⁵ сн ⁵ сн ⁵ он	н	н	н	CH3	осн3	CH3
	н	н	CH ² CH ² CH ² OCOCH ³	н	н	н	CH ₃	осн3	CH.3
0	н	н	CHSCHSCHSOCOCH3	н	H	н	CH3	осн ₃	CH ₃
	н	н	CH ² CH ⁵ CH ² H(CH ³) ⁵	н	н	н	CH3	OCH3	CH3
0	H	н	сн ² сн ² сн ⁵ н(сн ³) ⁵	H	н	н	CH ₃	осн3	CH3
									cont.

K	a ¹⁵	R ¹	R ²	R ³	R4	R ⁵	R ⁶	R ⁷	. R ²
5	н	н	CH2CH2CH2NHCOC2H5	н	н	. н	CH3	осн	Cit ₃
02	н	Н	CH2CH2CH2NHCOC2H5	н	н	н	CH3	ссн3	CH3
5	н	н	сн•сн-сосн ³	н	н	н	CH3	ссн3	CH3
0	н	н	сн•снсосн ₃	н	н	н	CH3	осн ₃	CH ³
;	н	H	CH2CH2COCH3	н	H	H	CH3	осн	CH3
0	н	н	CH2CH2COCH3	н	H	н	CH3	осн ₃	CH ₃
	н	н	CH•CH ← (○)	н	н	н	CH3	осн3	CH ₃
٥	н	H	CH•CH—	н	н	н	CH3	осн ₃	CH ₃
	H	н	CH2CH2-	н	H	н	CH3	осн ₃	CH3
0	н	H	CH ² CH ² -	н	н	н	CH3	осн ₃	CH ₃
	H	CH3	н	` сн ₃	н	н	CH3	осн ₂ сн-сн ₂	CH3
0	н	CH3	н	CH3	н	н .	CH3	осн ₂ сн-сн ₂	CH ₃
	н	H	CH ² -(O)	н	H	H	CH ³	асн ₃	снз
0	H	H		н	н	н	CH3	асн ₃	СНЭ
	H	н	o🔘	H	н	н	CH3	осн ₃	CH3
0	Ħ	H	۰-◎ _	н	н	н	CH3	осн ₃	CH3
	н	H	OCH ² CH ²	H	н	н	CH3	осн3	CH3
0	H	H	OCH ₂ CH ₂ ·	H	н	н	CH ³	осн3	CH3

	R ¹⁵	R ¹	R ² ↔	R ³	R ⁴	R ⁵	R ⁶	R ⁷	8 8
	н	н	осн ₂ сн	н	н	н	CH3	осн3	CH3
)	н	н	OCH ₂ CN	н	н	H	CH ₃	осн3	CH ³
	H	H	OCH2COOC2H5	н	H	H	CH ³	осн3	CH ₃
	н	н	och ₂ cocc ₂ H ₅	H	н	н	CH ₃	CCH3	сн ₃
	н	H	OCH ₂ CH ₂ OH	н	н	н	CH3	осн3	CH ₃
	н	H	OCH2CH2OH	н.	H	н	CH3	осн3	CH3
	н	н	OCH2CH2OCOCH2-(O)	н	н.	н	CH ₃	ocii3	CH ₃
	н	н	осн ₂ сн ₂ ососн ₂ (С)	н	н	н	СН ³	_{осн} 3	CH ³
	н	H	OCH2CH2XH2	н	H	H	CH3	осн3	CH ₃
	н	н	OCH ² CH ² NH ²	н	н	Ħ	CH3	OCH3	CH3
	н	н	осн ⁵ сн ⁵ инсосн(сн ³) ⁵	н	н	H	CH ³	осн ₃	CH ₃
	н	H	OCH2CH2NHCOCH(CH3)2	н	H	H	CH3	OCH3	CH ₃
	н	H	OCH ₂ CO	. н	н	H	CH3	och ³	CH ₃
	H	H	осн ₂ со —(О́́́́)	н	H	н	CH3	осн3	CH ₃
	н	н	` ∞ - ⊘ ¯	н	н	H	CH3	OCH3	CH3
	н	H	∞ -⊘ .	н	н	н	CH3	осн3	CH ³
	н	н	Ca(CH ⁵)³0←	н	н	н	CH ₃	OCH3	CH ³
	н	н	CO(CH ₂)30-O	н	н	н	CH ₃	осн ₃	CH3
	н	н	>	н	, н	H	CH3	осн3	CH3
•									con

X	R ¹⁵	я¹.	R ²	A ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸
o	н	н		. н	н	н	CH3	OCH 3	СН
	н	н	CCOCH ⁵ CH ⁵ OCH ³	CH ₃	н	н	CH ₃	осн ₃	CH.
)	н	H	COOCH ² CH ² OCH ³	CH3	н	н	CH3	осн ₃	. сн
	. н	н	COOCH ² —	CH3	н	H	сяз	осн ₃	CH ²
	н	н	соосн ₂ —	CH3	н	н	CH3	осн ₃	CH
	н	Н	CH ₂ OH	CH ₃	к	H	CH ₃	осн ₃	CH3
)	н	н	CH ² OH	CH3	н	H	CH ₃	осн ₃	CH ²
	Н	Н	СН ₂ ОСО — (СНЗ	н	н	CH ₃	осн ₃	CH3
1	Н	н	CH ² 0C0(O)	. сн3	н	н	CH ₃	осн ₃	CH3
	H	H	соосн3	CH3	н	H	CH3	OCH2CH=CH2	CH ³
	н	н	COOCH ³	CH3	H	H	CH3	OCH2CH=CH2	CH ₃
	H	H	CH2CH2OCH3	H	H	H	CH ₃	0СН ₃	CH3
	н	H	сн ₂ сн ₂ осн ₃	H	н	H	CH ₃	OCH ₃	CH3
	H	H	CH(CH ³) ⁵	н	H	H	CH3	OCH2CH=CH2	CH ³
ł	н	H	CH(CH ₃) ₂	н	н	н	CH3	OCH2CH=CH2	CH3
	н	Н	с(сн3)3	н	H	н	CH3	OCH2CH=CH2	CH ³
	H	н	с(сн ₃)3	н	н	н	CH3	OCH ² CH=CH ²	CH ³
	H	CH3	осн3	CH3	н	н	CH ₃	осн ₃	CH3
									con

cont.

x	R ¹⁵	R ^T	R ²	R ³	R ⁴	R ^S	R ⁶	R ⁷	R ⁸
so	H	снз	осн ³	CH3	н	н	CH3	0СН ₃	CH ³
S	н	CH ₃	осн3	CH ₃	н	н	CH ₃	CH ₃	н
50	н	CH3	осн3	CH3	н	н	CH ₃	CH3	н
S	К	CH3	och ₂ ch ₂ och ₃	CH ₃	H	н	CH3	осн ₃	CH ³
02	H	CH3	och ₂ ch ₂ och ₃	CH ₃	н	н	CH3	осн ₃	CH ₃
5	н	CH3	och ₂ ch ₂ och ₃	CH3	н	H	н	снз	CH3
0	H	CH ₃	OCH2CH2OCH3	СНЗ	н	H	н	CH ³	CH3
•	H	CH3	сосн3	CH ₃	н	н	CH3	OCH3	CH ₃
0	н	CH3	сосн	CH3	Н,	н	CH ₃	осн3	CH ³
	н	CH3	сосн3	CH3	н	н	CH ₃	н	CH3
0	н	CH ₃	COCH ³	CH3	н	н	CH ₃	н .	СНЗ
•	H	CH3	COC ₂ H ₅	CH3	H	н .	CH3	осн ₃	ᅄ
0	н	CH ₃	COC2H5	CH ₃	H	н	CH ³	осн ₃	CH3
i	CH.3	CH3	CH ³	CH ₃	H	н	CH ₃	осн ₃	CH ₃
0	CH ³	CH3	сж ₃	CH ₃	н	H	CH ₃	осн ₃	CH ³
•	H	CH3	CH ₃	CH3	H	н	CH ₃	СН3	CH ³
0	H	CH3	CH ³	СНЭ	н	н	CH ₃	снз	сн ₃
•	н	сн3	C2H5	CH3	н	н	CH3	осн3	CH ₃

-čant.

R ¹⁵	R ¹	я ²	· R3	R ⁴	R ⁵	R ⁶	R ⁷	. 88
э н	CH3	C ₂ H ₅	СНЭ	н	н	CH3	осн ₃	СН
н	CH3	с ₂ н ₅	CH3	н	н	CH ₃	осн3	н
) н	CH3	C2H5	сн3	н	н	CH3	осн ₃	н
H	CH3	CH(CH ₃)2	сн ³	н	н	CH ₃	осн3	снэ
н (CH3	сн(сн ³) ⁵	снз	н	н	CH3	осн3	CH ³
н	CH3	сн(сн ₃)2	CH3	н	н	CH3	СНЗ	CH3
) н	CH3	сн(сн ₃)2	CH3	н	н	CH ₃	сн ₃	CH ³
H	CH ₃	COCH ₂ -O	сн ₃	н	н	CH ₃	осн ₃	CH ₃
н	CH ₃	COCH2-(C)	CH ³	н	н	CH ³	осн3	CH3
н	осн3	Br	осн3	н	н	CH ₃	осн3	CH ₃
н	0CH ₃	Br	осн3	H	н	CH ₃	0СН3	сн ₃
н	осн ₃	8r	осна	н	H	CH ₃	СНЗ	н
H	осн ₃	8r	асн3	н	H	CH ₃	CH ₃	H
H	с ₂ н ₅	CH	c _z ĸ _s	н	H	CH3	осн3	CH ₃
н	c ₂ н ₅	CN	C ₂ H ₅	н	H	CH3	оснз	CH ³
H	с ₂ н ₅	CN	c ^{SH2}	н	H	CH3	oc ₂ #5	. сн3
н	c _z H ₅	CN	c ₂ H ₅	н	H	CH3	0C2H5	CH3
н	CH ³	осн ₃	CH3	CH3	H	сн3	осн ₃	CH ³
								cont.

cont.

									
K	R ¹⁵	R ¹	R ²	R ₂	R ⁴	R ⁵	R ⁶	R ⁷	R ^S
50	H	CH ₃	CCH ³	сн ₃	CH3	н .	CH3	осн3	СН3
s	H	снэ	OCH3	н	CH3	H	CH ₃	OCH 3	CH ₃
50	н	CH ₃	осн3	н	CH3	H	CH ³	осн ₃	CH3
•	н	Cl	осн3	н	OCH ³	н	. сн ₃	осн3	CH3
0	н	Cl	осн ₃	н	.осн3	·H	CH ₃	och ³	CH3
S	н	C1	C1	C1	H	н	CH3	осн ₃	CH3
0	н	C1	C1	· C1	н	H	CH3	OCH3	CH3
;	н	C1	C1	cı	н	H	CH3	OCH ² CH-CH ²	CH3
0	н	C1	CI	C1	H	H	CH3	och ^S cH∗cH ^S	CH3
;	н	Cl	C1	C1	c1	H	CH3	och3	CH ₃
0	н	Cl	C1	C1	C1	н	CH3	och3	CH3
;	н	C1	ct	C1	C1	н	CH3	OCH ² CH=CH ²	CH ₃
0	H	c1	c 1	C1	C1	н	CH3	OCH ² CH=CH ²	CH3
5	н	сснз	Br	н	оснз	н	CH3	осн ₃	CH ³
50	$\mathbf{H}_{\mathbf{k}^{-1}}$	осна	Br	н	осн 3	Iŧ	CH3	CCH3	CH ₃
5	н	осн3	c1	C1	0C2H5	H	CH3	осн ₃	CH3
0	H	осн3	c 1 .	C1	^{0С} 2 ^Н 5	H	CH3	осн ³	снз
;	н	OCH3	cı	C1	0C2H5	н	CH ₃	CH3	н

X	R ¹⁵	R ¹	a ²	R ³	R ⁴	R ^S	a ⁶	R ⁷	88
so	н	осн3	· c1	Cl	GC2H5	Н	сн ₃	СНЗ	н
s	н	сосн3	CH3	CH3	CH3	н	CH3	осн ₃ .	CH ₃
so	H	соснз	снз	CH ₃	CH3	н	CH3	осн3	CH ₃
5	H	F .	C1	н	£1	н	CH3	осн	CH ₃
0	H	F	C1	н	C1	H	CH ₃	осна	CH3
	н	C1	сн ₂ соосн ₃	C1	Ħ	н	CH ₃	осн3	CH3
0	H	C1	сн ² соосн ³	C1	Ħ	н	CH3	оснз	CH ₃
	H	C1	CH ₂ CN	C1	н	н	CH ³	осн3	CH ³
0	н	C3	CH ₂ CN	C1	H	H	CH3	осн ₃	CH ₃
3	н	-CH=CH-		-сн=сн	-сн=сн-	H	сн3	осн ₃	CH ³
	н	н	CCH CH3		н	н	сн ₃	осн3	снз
)	н	H	COH CH3	н	н	н	CH ₃	асна	сн ₃
	н	н	- ◎	н	н	н	CH ₃	осн ₃	CH3
3	Н	H	- ◎	H	н	H	CH ₃	OCH3	CH ³
	н	н	-0CH ₂ 0-		н	н	CH ₃	осн3	сн ₃

١	R ¹⁵	R ¹ R ²		R ³	R ⁴	я ⁵	R ⁶	R ⁷	R ⁸
0	н	н	-осн ₂ о-		н	н	CH ₃	OCH 3	CH ₃
;	H	н	-осн ₂ о-		н	н	CH3	CH ₃	CH ₃
0	н	н	-0CH ² 0-	=	н	н	CH3	сн3	CH3
	H	н	~~. ^-		н	н	ск3	осн ₃	CH ₃
0	н	н	- ₀ × ₀₋	•	н	н	СНЗ	осн ₃	СН.3
	н	-CH=CH-CH=N-		Н	H	H	CH3	осн _з	CH ₃
0	H	-CH=CH-CH=X-		н	н	н	CH ₃	0CH ₃	CH ₃
	н	-CH=CH-CH=CH-		н	H	н	CH ₃	осн ₃	CH3
0	н	-CH=CH-CH=CH-		H	н	н	CH ₃	OCH ₃	СН ₃
	н	н	-CH=CH-CH=CH-		н	н	CH ₃	осн ₃	сн3
)	H	н	-CH=CH-CH=CH-		н	н	CH3	осн ₃	CH ₃
	H	-сн ₂ сн ₂ сн ₂ сн ₂ с	-	н	H	H	CH ₃	осн ₃	CH ₃
0	н	-сн ₂ сн ₂ сн ₂ сн ₂	•	н	н	н	CH3	осн ³	CH ₃
	н	осн ₃	-сн ₂ сн ₂ сн ₂ -		C1	н	CH ₃	оснз	CH3
0	н	осн ₃	-сн ₂ сн ₂ сн ₂ -		c1	н	CH3	OCH 3	CH ₃
	н	осн3	-сн 2сн 2сн2-		· c 1	н	CH3	oc ₂ H _S	сн ₃

-- cont.

K	R ¹⁵	R ¹	R ²	R3	R ⁴	R ⁵	R ⁶	R ⁷	. _R S
٥	н	· 0CH3	-cH ₂ CH	CH ₂ -	¢1	Н	сн3	ос ₂ н ₅	CH ₃
	н	- C)	н = сн - сн = с∙-	CH2CH2 -	н	н	CH ₃	66H ³	CH ₃
0	H	- Ci	i = сн - сн = с -	сн ² сн ⁵ -	н	н	CH ³	осн ₃	сн ₃
	н	н	<u>©</u> ©	`co-	н	н	CH ₃	осн ₃	сн ³
0	н	н		`co-	н	н	CH3	осн ₃	CH ₃
	н	н	-осн ₂	0-	н	CO ₂ CH ₃	он ₃	OCH ³ .	CH ² .
9	н	Н	-061 ₂ 1)-	H	со _г сн _з	-	. осн ₃	ca ₃
	н	H	-OCH2		н	CO ₂ C ₂ H ₅	CH3	OCH ³	CH ³
)	H	н	-0CH ₂ C		н	CO ₂ C ₂ H ₅	CH ₃	осн ³	_
	H	Ĥ	-0CH ₂ (н	со ₂ с(сн ₃) ₃	CH ³	осн ₃	CH ³
	, н,	н	-0CH ₂ C		н	CO ² C(CH ³) ³		0CH ₃	CH ³
	H	H	-0CH ₂ C		H	CO2CH2-(C)	÷+ − '3 CH ₃	осн ³	сн ³
	н	н	-0CH ₂ 0	I -	н	(0 ₂ CH ₂ -(O)	CH ₃	OCH ³	
	H	н ·	-0CH ₂ 0		н	"√O	CH ³	OCH ³	CH ₃
	н	н	-0CH ₂ 0		н			-	CH ₃
	н	н	-0CH ₂ 0		н	CONH ₂	CH ³	OCH ³	CH ³
			2-		••	2	CH ³	OCH3	CH ₃

cont.

	R ¹⁵	R ¹	R ²		R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸
0	н	H		-0CH ₂ 0-		Н	CONH ₂	CH ₃	осн3	CH,
•	H	н		-осн ₂ о-		н	CONHC ₂ H ₅	CH ₃	OCH ³	CH.
0	Н	H		-осн ₂ о-		н .	CONHC ₂ H ₅	CH ³	OCH ₃	CH.
	н	н		-0CH ₂ 0-		н ''	CONHCH2-		осн ₃	CH.
0	н	н		-осн ₂ 0-		н	соннсн2(С	ў ан,	OCH3	CH.
	н	н		-0CH ₂ 0-		H	CONH -	CH ₃	осн ₃	CH.
)	Н	н		-осн ₂ о-		н	CONH-(O)	대3	0CH ₃	CH.
	H	н		-осн ₂ о-		н	CON (CH ₃) ₂	CH ₃	0CH ₃	CH.
)	н	H		-осн ₂ о-		н	CON(CH ₃)2	CH ₃	OCH3	CH.
	Н	CH ₃	CH3		CH3	H	н	CH3	OCH2CH2OCH3	CH.
)	н	CH ₃	CH3		CH ₃	н	H	CH ₃	OCH ₂ CH ₂ OCH ₃	CH.
	H	н	осн3		н	н	н	- CH	=CH-0-	н
)	н	н	och ₃		H	н	н	-CH	=CH-O-	н
	н	H	och3		н	. н	н	н	-0-CH=CI	1 -
)	H	H	OCH ₃		н	н	н	н	-0-CH+C	1-
	Н	H	осн ₃		н	н	H	-CH	*CH-NH-	H
	н	H	осн3		н	н	H	-СН	≈CH-NH-	н
	н	н	OCH ₃		Н	н	н	н	-NH-CH=0	:11-

x	R ¹⁵	R ¹ ·	R ²	٤3	R [‡]	R ⁵	R ⁶	R ²	28
SO	н	н	осн3	н	н	н	ä		-XH-CH+CH-
S	н	н	OCH3	н	н	н	-CH	•Сн-и(сн ₃)-	н.
so	н	H.	осн3	н	н	н		•CH-H(CH ₃)-	н.
S	H	н.	OCH ³	H	н	н	н		сн₃)-сн=сн-
0	H	н	OCH3	н	H	н	н		(CH3)-CH=CH-
•	H	CH3	CH _Z C*CH	CH3	н	н	Сн3	осн ₃	CH3
0	Н	CH3	CH2C=CH	CH ³	н	н	CH3	ССНЭ	CH ₃
1	H	н	CH2CH2CH2O-O	H	н	н	CH3	осн ₃	CH ₃
) 1	н	Н	ᅜᅥᆉᄱᆉᇝ	H	н	н	CH ₃	осн ₃	CH ₃
1	H	н	OCH2CH2CH2O-O	н	н	н	CH ₃	осн ₃	CH ₃
•	4	н	OCH ₂ CH ₂ CH ₂ O-	н	н	H	CH3	OCH3	СН ₃
	4	CH ₃	о(сн ₂)6сн3	CH ₃	н	H	CH ₃	осн ₃	CH ₃
) !	4	CH3	о(сн ²) ⁶ сн ³	CH ₃	H	н	CH ₃	осн ₃	CH ₃
1		н	C2H5	H	H	H	CH ₃	OCH2CH=CH2	CH ₃
H		н	C2H5	H	н	н	CH ₃	OCH2CH=CH2	CH3
Н		н	осн3	H	H	∞ -©	CH ₃	осн ₃	CH ₃
Н		н	н	осн ₃	H	:0 - ⊘	CH3	осн ₃	cH ³
н	i	н	ocH ³	н	н	u-Ō	CH3	осн3	он ₃
						_			cont

x -	R ¹⁵	RT	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸
0	н	н	Н	осн3	н	co-⊘ ć	CH ₃	осн ₃	сн3
5	н	н	CH3	CH ₂ OCO 🔘	н	co(C)	CH ₃	осн ₃	CH ₃
;	н	К	CH ² 0C0 -	снэ	н	co -🔘	CH3	осн ₃	CH ₃
•	H	н	-0EH ₂ 0-		н	COCZHS	ᅄ	осн ₃	CH3
0	н	H	-осн ₂ о-		н	coc2H2	CH3	осн ₃	CH ₃
0	н	н -	(C)	CH ³	H	соосн3	CH3	осн3	CH ³
	H	-oc´	© 	Н	н	H	СНЗ	осн3	сн3
0	н	-cc´	CO-	н	H	н	CH ₂	0СН ₃	CH ₁
	н	н	SCH ³	н	н	н	CH ₃	асн _э	CH ³
	н	н	CH(CH ³) ^S	H	н	н	CH3	OCH 2	CH3
٥	H	H	сн(сн ₃)2	ж .	н	н	снз	OCH Z	сн3
	н	н	сн ⁵ сн ⁵ сосн ³	н	н	н	снз	осн ₂ сн•сн ₂	СНЗ
0	н	н	CH ^S CH ^S COCH ³	н	H	н	CH3	OCH ₂ CH=CH ₂	CH ³
0	н	н	CH3	CH3	н	соос(ся3)3	CH3	OCH3	сн3

-- Table 1 cont.

x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷ .	R ⁹
s	н	н	СНЗ	CH3	н	CON(CH3)2	CH3	осн3	сн3
SO	H	н	снз	CH3	н	CON(CH3)2	СНЗ	осн ₃	CH ₃
S	н	н	8r	н	н	н	CH3	OCH2CH+CH2	CH3
SO	н	н	8r	н	н	н	СНЗ	осн ₂ сн=сн ₂	CH3
S	н	CH ₃	сн3	CH ₃	н	н	CH ₃	сн3	н
50	н	снз	сн ₃	CH ₃	н	н	CH3	сн ₃	н :
s	н	CH ₃	CH ³	CH ₃	н	н	н	CH ₃	CH3
SO	н	CH ₃	CH3	сн3	H	н	н	сн3	CH ²
S	н	CH ₃	сн3	сн ₃	H	н	CH3 -	н	CH3
50	н	CH ₃	сн3	сн3	н	н	CH3	н	CH3
S	н	CH3	сн3	н	CH3	н	CH3	CH ₃	н
S 0	н	CH ₃	сн3	н	CH3	н	CH3	сн _з	н
S	н	CH3	CN	CH3	н	н	CH3	ос ₂ н ₅	CH ₃
50	н	CH3	CN	CH3	H	н	CH ₃	ос ₂ н ₅	CH3
so	н	н	соосн3	сн ₃	H	H	н	осн3	C2H5
s	н	н .	-сн ₂ сн ₂	CH ₂ -	н	н	CH3	осн3	CH3

Table 1 cont.

x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	_R 8
so	н	н	-сн ₂ сн ₂ сн ₂ -	• .	н	н	снз	осн3	сн ₃
50	H	Н	осн ₃	н	н	н	-CH ₂ C	н ₂ сн ₂ 0-	н
SO.	н	н	оснз	н	н	н .	н	-осн ₂ сн ₂ -	
S	н	н.	SOCH3	н	н	н	снз	осн3	сн3
50	H	н	SOCH3	H	н	н	СНЗ	осн ₃	CH ₃
S	н	н	сн ₃	сн ₃	н	н	сн3	-осн ₂	сн3
so	н	н	сн ₃	снз	н	н	CH3	-осн ₂ -	сн ₃
S	н	-CH	=CH-CH=CH-	-Сн≠Сн	-CH=CH-	н	CH3	осн3	CH3
90	н	н	^{NO} 2	. н	н	н	CH3	осн3	сн3
s	н	н	CF ₃	н	н	н	сн3	0CH2-0	сн ₃
\$0	н	н	CF ₃	н	н	н	CH ₃	OCH 2	CH ₃
s	н	н	сн ₂ сн ₂ соос ₂ н ₅	н	н	. н	снз	осн3	. сн _з
so	н	H	оснз	н	н	о С-ос(сн ₃)3	снз	осн _з	СНЗ
so	н	н	CH ₃	сн ₃	н	H	H	осн ₃	с ₂ н ₅

The invention takes into consideration that compounds that structurally deviate from the formula I, after administration to a living organism may be transformed to a compound of formula I and in this structural form exert their effect. Such compounds structurally deviating from compounds of the formula I, are included in the scope of the invention.

Likewise, certain compounds of formula I may be metabolized into other compounds of formula I

10 before exerting their effect. Compounds of the invention wherein X is S are thus believed to exert their antisecretory and cytoprotective activities after metabolism to compounds wherein X is SO and compounds of the invention wherein R⁵ is R¹⁴CO are believed to exert antisecretory and cytoprotective activity after metabolism to compounds wherein R⁵ is

H. These considerations are also a further aspect of the invention.

Further, it is believed that all compounds of
formula I wherein X is SO after administration to a
living organism, exert their antisecretory and cytoprotective effects after metabolic or pure chemical
transformation to another, reactive species. Accordingly, the same is true also for the compounds of

25 formula I wherein X is S, but via initial transformation to the corresponding compounds of formula I wherein X is SO. These considerations as well as such reactive species per se are included within the scope of the present invention.

30 Preparation

Compounds of formula I above may be prepared according to the following methods:

a) -Oxidizing a compound of the formula I,

$$R^{\frac{3}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}}$$

$$R^{\frac{3}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}}$$

$$R^{\frac{3}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}}$$

$$R^{\frac{3}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}}$$

wherein X is S and R¹⁵, R¹, R², R³, R⁴, R⁵, R⁶, R⁷ and R⁸ have the meanings given, to give a compound of the same formula I wherein X is SO. This oxidation may 5 be carried out by using an oxidizing agent selected from the group consisting of nitric acid, hydrogen peroxide, peracids, peresters, ozone, dinitrogentetraoxide, iodosobenzene, N-halosuccinimide, I-chlorobenzotriazole, t-butylhypochlorite, diazabicyclo-10 [2,2,2] - octane bromine complex, sodium metaperiodate, selenium dioxide, manganese dioxide, chromic acid, cericammonium nitrate, bromine, chlorine, and sulfuryl chloride. The oxidation usually takes place in a solvent wherein the oxidizing agent is present in some excess in relation to the product to be oxidized.

The oxidation may also be carried out enzymatically by using an oxidating anzyme or microbiotically by using a suitable microorganism.

20 b) Reacting a compound of the formula

$$R^{2}$$

$$R^{3}$$

$$R^{3}$$

$$R^{4}$$

$$R^{5}$$

$$R^{5}$$

$$R^{1}$$

$$R^{1}$$

$$R^{2}$$

$$R^{1}$$

$$R^{2}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{4}$$

$$R^{5}$$

$$R^{5}$$

with a compound of the formula

in which formulas R¹⁵, R¹, R², R³, R⁴, R⁵, R⁶, R⁷ and R⁸ are as defined previously and wherein one of Z¹ and Z² is SH and the other is a leaving group, gives a compound of the formula I wherein X is S.

Examples of leaving groups Z¹ and Z² in the compounds II and III are halogens, preferably chlorine, bromine or iodinem acyloxy radicals, for example residues of strong organic sulfonic acids, for instance of an arylsufonic acid, for example tosyloxy or an alkylsulfonic acid, for example mesyloxy, alkylmercapto groups, for example methylmercapto, alkylsulfinyl groups, for example methylsulfinyl and the like.

Thus, Z¹ or Z² when designating leaving groups may be a reactive esterified hydroxy group. The esterification may be carried out with an organic acid or with an inorganic acid such as HCI, HBr or H₂SO₄.

The reaction of a compound of formula II above
with a compound of formula III is conveniently carried
out in the presence of a suitable solvent that is inert
under the reaction conditions utilized as described
hereinafter. The reaction may further be carried out in
the presence of a suitable base. Suitable bases
include, for example, inorganic bases such as sodium

or potassium hydroxide, sodium or potassium alkoxide, sodium or potassium hydride and the like, organic bases such as tertiary amines, for example triethylamine and the like:

Suitable solvents for the above described reaction include, for example, alcohols, preferably lower alkanols such as methanol and ethanol, mixtures of such alcohols with water, ethers, such as tetrahydrofuran, halogenated hydrocarbons, such as methylene chloride. Aprotic solvents such as ethers and halogenated carbons are necessary in the case of sodium and potassium hydride.

The reaction of the compounds of formulas II and III may be carried out at a temperature between the
60 ambient temperature and the boiling temperature of the reaction mixture. It is preferred to carry out the reaction, however, at a temperature at or close to the boiling point of the reaction mixture for the preparation of a compound of the formula I wherein R⁵ is H.

c) Esterification of a compound of the formula

$$R^{\frac{5}{2}} \xrightarrow{R^{\frac{7}{2}}} R^{\frac{5}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}} \xrightarrow{R^{\frac{1}{2}}}} R^{\frac{1}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}} \xrightarrow{R^{\frac{1}{2}}}} R^{\frac{1}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}}} \xrightarrow{R^$$

wherein R¹⁵, R⁵, R⁶, R⁷ and R⁸ are as defined above and Y¹, Y², Y³ and Y⁴ represent either R¹, R², R³ and R⁴ according to the above definition, respectively, or the groups (Z)_n-A-COOH, COOH and (Z)_n-A-OH, whereby Z, n and A are as defined above, by reaction with the appropriate alcohol R⁹OH, R¹⁰OH or carboxylic acid R¹⁰COOH, respectively, to the formation of a compound of formula I containing a radical R¹, R², R³ and/or R⁴ which is either of the ester groups 75 (Z)_n-A-COOR⁹, COOR¹⁰ or (Z)_n-A-OCOR¹⁰.

The esterfication is carried out as an ordinary esterfication, in the presence of an acid catalyst such as sulfuric acid, hydrochloric acid and p-toluenesulphonic acid and, if necessary, in the presence of an inert solvent such as toluene.

d) Acylation of a compound of the formula

$$R^{\frac{1}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}} \xrightarrow{R^{\frac{1}{2}}} R^{\frac{1}{2}}$$

wherein R¹⁵, X, R¹, R², R³, R⁴, R⁶, R⁷ and R⁸ are as defined above, by reaction with an appropriate acylating agent (R¹⁴CO)₂O, R¹⁴COX¹, whereby X¹ is a leaving group such as C1, N₃ and p-nitrophenoxy, R³NCO, whereby R³ is defined by the relation R³NH equals R¹⁴, provided that R³ is K when R¹⁴ is amino, to the formation of a compound of formula I wherein R⁵ is R¹⁴CO as defined above.

The acylation is preferably carried out in the presence of a base such as triethylamine, K_2CO_3 and NaOH and with a solvent such as tetrahydrofuran, acetonitrile and water. Normally, if the benzimidazole moiety is asymetrically substituted, both the N(1)-

90

20

and the N(3)-acyl derivatives are obtained, and therefore, if necessary, the two components have to be separated. This may be done by recrystallizations or by extractive or chromatographic techniques.

e) Hydrolyzing a compound of the formula

$$R^{5} \xrightarrow{R^{7}} R^{6} \xrightarrow{R^{1}} R^{2}$$

$$\downarrow_{i_{1}} R^{1} \xrightarrow{\downarrow_{i_{1}} R^{1}} R^{2}$$

$$\downarrow_{i_{1}} R^{1} \xrightarrow{\downarrow_{i_{1}} R^{1}} R^{2}$$

$$\downarrow_{i_{1}} R^{2} \xrightarrow{\downarrow_{i_{1}} R^{1}} R^{2}$$

$$\downarrow_{i_{1}} R^{2} \xrightarrow{\downarrow_{i_{1}} R^{2}} R^{3}$$

wherein X, R15, R1, R2, R3, R4, R6, R7 and R8 are as defined above and Z3 is a suitable N-protecting group such as alkanoyl, carboalkoxy and trimethylsilyl, to the formation of a compound of the formula I wherein 10 R⁵isH.

The alkanoyl group in Z³ can have 1-6 carbon atoms and the carboalkoxy group 2-6 carbon atoms. The hydrolysis may be performed in alkaline solution or in acidic solution, the latter mainly for compounds 15 wherein X is S;

whereafter the compound of the formula I obtained if desired, when X is -S-, is converted to a physiologically acceptable salt or oxidized to form a compound of the formula I wherein X is -SO-.

Depending on the process conditions and the starting materials, the end products of the formula l wherein X is S is obtained either as the free base or as a salt. The end products of the formula I wherein X is -SO- are obtained as the free base. Both the free base 25 and the salts of these end products are included within the scope of the invention. Thus, basic, neutral or mixed salts may be obtained as well as hemi, mono, sesqui or polyhydrates. Acid addition salts of the new sulficides may in a manner known perse be 30 transformed into free base using basic agents such as alkali or by ion exchange. The free bases of the sulfides obtained may also form salts with organic or inorganic acids. In the preparation of acid addition salts preferably such acids are used which form 35 suitable therapeutically acceptable salts.

Examples of such acids are hydrohalogen acids, sulfonic acid, phosphoric acid, nitric acid, and perchloric acid; aliphatic, alicyclic, aromatic or heterocyclic carboxyl or sulfonic acids, such as formic acid, 40 acetic acid, propionic acid, succinic acid, glycolic acid, lactic acid, malic acid, tartaric acid, citric acid, ascorbic acid, maleic acid, hydroxymaleic acid, pyruvic acid, phenylacetic acid, benzoic acid, p-aminobenzoic acid, p - hydroxybenzoic acid, salicyclic 45 acid or p-aminosalicylic acid, ambonic acid, methanesulfonic acid, ethanesulfonic acid, hydroxyethanesulfonic acid, ethylenesulfonic acid, halogenbenzenesulfonic acid, toluenesulfonic acid, naphtylsulfonic acid or sulfanilic acids, methionine, 50 tryptophane, lysine or arginine.

These or other salts of the new sulfide compounds, as e.g. picrates, may serve as purifying agents of the free bases obtained. Salts of the bases may be formed, separated from solution, and then the free 55 base can be recovered in higher purity from a new salt solution.

Racemates obtained can be separated according to

known methods, e.g. recrystallization from an optically active solvent, use of microorganisms, reactions 60 with optically active acids forming diastereomeric salts which can be separated, (e.g. separation based on different solubilities of the diastereomers), acylation of the benzimidazole nitrogen ($R^5 = H$) or another nitrogen or oxygen atom in a substituent by an 65 optically active activated carboxylic acid (e.g. acid chloride), followed by chromatographic separation and deacylation.

Suitable optically active acids for salt formation are the L- and D-forms of tartaric acid, di - o - tolyl - tartaric 70 acid, malic acid, mandelic acid, camphorsulfonic acid or quinic acid, and for acylation O - methylmandelic acid. Preferably the more active part of the two antipodes is isolated.

In the case of diastereomeric mixtures (racemate 75 mixtures) these may be separated into stereoisomeric (diastereomeric) pure racemates by means of chromatography or fractional crystalliza-

The starting materials utilized in the processes a 80 and c-e are obtained from the process b. The starting materials used for process b are in some cases known, but in most cases unknown. These unknown starting materials may, however, be obtained according to processes known perse.

85 Starting materials of the formula!

90

$$R^{2}$$

$$R^{3}$$

$$R^{3}$$

$$R^{4}$$

$$R^{5}$$

$$R^{5}$$

$$R^{5}$$

wherein Z1 is SH may be obtained from the corresponding o - phenylenediamine by reaction with potassium ethylxanthate (Org. Synth. Vol. 30, p. 56) orthiophosgene.

The compounds of the formula II wherein Z¹ is alkylmercapto and alkylsulfinyl may be obtained from the above mentioned compound by simple S alkylation with alkyl halide and by oxidation of the product from the S - alkylation, respectively.

95 The compounds of the formula II wherein Z¹ is halogen or acyloxy may be obtained from compounds of the same formula wherein Z^1 is OH by treatment with POCl3, POBr3 and the like or the appropriate acyl halide, respectively. The starting 100 material wherein Z1 is OH is obtained from the corresponding o - phenylenediamine by reaction with phosgene.

The o-phenylenediamines required may be obtained from the corresponding substituted ben-105 zenes according to processes known perse, e.g. by the consecutive processes: nitration, reduction, acetylation, nitration, deacetylation and reduction, or from one of the intermediary stages just mentioned. In order to obtain a o - phenylenediamine wherein R5 110 is other than H, acylation (by the group R¹⁴CO) is preferably made on the nitro - aniline stage. Starting materials of the formula

wherein R¹⁵ is H, may be obtained either from the correspondingly substituted (R⁶, R⁷ and R⁸) 2 - methyl - substituted pyridine N - oxide via a known rearrangement to the intermediate 2 - pyridinylmethanol or via a hydroxymethylation of the substituted (R⁶, R⁷ and R⁸) pyridine to give the same intermediate, and then treatment of the 2 - pyridinylmethanol with halogenating agents such as thionyl chloride or O - acylating agents such as p - toluenesulfonyl chloride to give compounds of the formula III wherein Z² is halogen and sulfonyloxy groups, respectively.

These leaving groups may then be substituted for alkylmercapto groups by treatment with e.g. sodium alkylmercaptide, which may then be oxidized to an 15 alkylsulfinyl group, or substituted for SH by treatment with e.g. NaSH.

For the preparation of intermediates of formula

wherein R⁷ is alkoxy, alkenyloxy, alkynyloxy, alkoxy-alkoxy and dialkylaminoalkoxy, a compound of
formula VII, wherein R⁷ is NO₂, is reacted by the corresponding sodium alkoxide. Analogously, for the preparation of an intermediate of formula VII wherein R⁶ and R⁷ or R⁷ and R⁸ form a ring structure including an oxygen atom at position 4, a compound of formula
VII wherein R⁷ is NO₂ and R⁶ or R⁸ represents hydroxyalkyl is reacted with a non-nucleophilic base.

The following intermediates A) and B) are included in the scope of the invention:

A) New compounds of the formula

- 30 wherein R^{1a} , R^{2a} , R^{3a} and R^{4a} are the same or different and selected from the groups
 - (a) H,
 - (b) alkyl containing 1-6 carbon atoms, including cycloalkyl,
- 35 (c) alkoxyalkyl containing 1-3 carbon atoms in the alkoxy part and 1-6 carbon atoms in the alkyl part,
 - (d) aryloxyalkyl containing 1-6 carbon atoms in the alkyl part,
- (e) arylalkyl containing 1-6 carbon atoms in the 40 alkyl part,
 - (f) aryl,
 - (g) alkoxy containing 1-6 carbon atoms,
- (h) alkoxyalkoxy containing 1-3 carbon atoms in the outer part and 1-6 carbon atoms in the part 45 nearest the aromatic ring,

- (i) aryloxyalkoxy containing 1-6 carbon atoms in the alkoxy part,
- (j) arylalkoxy containing 1-6 carbon atoms in the alkoxy part and
- 50 (k) aryloxy,
 - R^{5a} is
 - (a) H,
 - (b) alkoxycarbonyl containing 1-4 carbon atoms in the alkoxy part,
- (c) arylalkoxycarbonyl containing 1-2 carbon atoms in the alkoxy part,
 - (d) dialkylaminocarbonyl containing 1-4 carbon atoms in each alkyl group, or
 - (e) arylaminocarbonyl,
- 60 and Z^{1a} is
 - (a) SH,
 - (b) ClorBr

and provided that not more than one of R^{1a}, R^{2a}, R^{3a} and R^{4a} is H, are suitable intermediates for the preparation of compounds of the formula I with R¹, R², R³, R⁴ and R⁵ having the same meaning as R^{1a}, R^{2a}, R^{3a}, R^{4a} and R^{5a}, respectively, according to method b.

B) New compounds of the formula

wherein R^{6a} and R^{8a} are

(a) Hor

70

90

- (b) alkyl containing 1-5 carbon atoms, and R7a is
- (a) alkenyloxy containing 2-5 carbon atoms, or
- (b) alkynyloxy containing 2-5 carbon atoms,
- (c) oxacycloalkyl containing one oxygen atom and 75 3-7 carbon atoms
 - (d) oxacycloalkoxy containing two oxygen atoms and 4-7 carbon atoms
 - (e) oxacycloalkylalkyl containing one oxygen atom and 4-7 carbon atoms
- 30 (f) oxacycloalkylalkoxy containing two oxygen atoms and 4-6 carbon atoms,
- (g) R^{6a} and R^{7a}, or R^{7a} and R^{8a} together with the adjacent carbon atoms in the pyridine ring form a ring wherein the part constituted by R^{6a} and R^{7a} or R^{7a} and R^{8a} is
 - -CH=CH-CH=CH-
 - ---O---(CH₂)_{pa}---
 - —CH₂—(CH_{)pa}—
 - -0-CH=CH-
 - wherein pa is 2, 3 or 4 and the O atom always is attached to position R^{7a} , and Z^{2a} is
 - (a) SH,
- 95 (b) halogen CI, Br, I or
 - (c) OH

and provided that not more than one of R^{6a} and R^{8a} is H, are suitable intermediates for the preparation of compounds of the formula I with R⁶, R⁷ and R⁸ having 100 the same meaning as R^{6a}, R^{7a} and R^{8a}, respectively,

according to method b.

For clinical use the compounds of the invention are formulated into pharmaceutical formulations for oral, rectal, parenteral or other mode of administration.

The pharmaceutical formulation contains a compound of the invention in combination with a pharmaceutically acceptable carrier. The carrier may be in the form of a solid, semi-solid or liquid diluent, or a capsule. These pharmaceutical preparations are a further object of the invention. Usually the amount of active compounds is between 0.1-95% by weight of the preparation, between 0.2-20% by weight in preparations for parenteral use and between 1 and 50

10 % by weight in preparations for oral administration. In the preparation of pharmaceutical formulations containing a compound of the present invention in the form of dosage units for oral administration the compound selected may be mixed with a solid, 15 powdered carrer, such as lactose, saccharose, sorbitol, mannitol, starch, amylopectin, cellulose derivatives, gelatin, or another suitable carrier, as well as with lubricating agents such as magnesium stearate, calcium stearate, sodium steryl fumarate and 20 polyethylene glycol waxes. The mixture is then processed into granules or pressed into tablets. Since the sulfoxides of the invention are susceptible to degradation in acid to neutral media, granules and tablets containing sulfoxides are preferably coated 25 with an enteric coating which protects the active compound from acid degraduation as long as the dosage form remains in the stomach. The enteric coating is chosen among pharmaceutically acceptable enteric-coating materials e.g. beeswax, shellac 30 or anionic film-forming polymers such as cellulose acetate phthalate, hydroxypropylmethylcellulose phthalate, partly methyl esterified methacrylic acid polymers and the like, if preferred in combination with a suitable plasticizer. To this coating various 35 dyes may be added in order to distinguish among tablets or granules with different active compounds or with different amounts of the active compound

Soft gelatine capsules may be prepared with
capsules containing a mixture of the active compound or compounds of the invention, vegetable oil, fat, or other suitable vehicle for soft gelatine capsules. Soft gelatine capsules may also be enteric coated as described above. Hard gelatine capsules may contain granules or enteric-coated granules of the active compound. Hard gelatine capsules may also contain the active compound in combination with a solid powdered carrier such as lactose, saccharose, sorbitol, mannitol, potato starch, corn starch, amylopectin, cellulose derivatives or gelatine. The hard gelatine capsules may be enteric coated as described above.

present.

Dosage units for rectal administration may be prepared in the form of suppositories which contain the active substance mixed with a neutral fat base, or they may be prepared in the form of a gelatine rectal capsule which contains the active substance in a mixture with a vegatable oil, paraffin oil or other suitable vehicle for gelatine rectal capsules, or they may be prepared in the form of a ready-made micro enema, or they may be prepared in the form of a dry micro enema formulation to be reconstituted in a suitable solvent just prior to administration.

Liquid preparations for oral administration may be prepared in the form of syrups or suspensions, e.g. 65 solutions or suspensions containing from 0.2 % to 20

% by weight of the active ingredient and the remainder constisting of sugar or sugaralcohols and a mixture of ethanol, water, glycerol, propylene glycol and polyethylene glycol. If desired, such liquid preparations may contain colouring agents, flavouring agents, saccharine and carboxymethyl cellulose or other thickening agent. Liquid preparations for oral administration may also be prepared in the form of a dry powder to be reconstituted with a suitable solvent prior to use.

Solutions for parenteral administration may be prepared as a solution of a compound of the invention in a pharmaceutically acceptable solvent, preferably in a concentration from 0.1 % to 10 % by weight. These solutions may also contain stabilizing agents and/or buffering agents and may be manufactured in different unit dose ampoules or vials. Solutions for parenteral administration may also be prepared as a dry preparation to be reconstituted with a suitable solvent extenporaneously before use.

The typical daily dose of the active substance varies within a wide range and will depend on various factors such as for example the individual requirement of each patient, the route of administration and the disease. In general, oral and parenteral dosages will be in the range of 5 to 500 mg per day of active substance.

The invention is illustrated by the following examples.

95 Example 1. Method a. Preparation of 4,6 - dimethyl - 5 - methoxy - 2 - [[(3,4 - dimethyl - 2 - pyridinyl) methyl] sulfinyl] - 1H - benzimidazole.

m-Chloroperbenzoic acid, 91% (0.53 g. 0.0028 mol) dissolved in CH^2Cl^2 (25 ml) and cooled to $-10^{\circ}C$ was 100 added under stirring to 4,6 - dimethyl - 5 - methoxy - 2 - [[(3,4 - dimethyl - 2 - pyridinyl) methyl] thio] - 1*H* - benzimidazole (0.91 g, 0.0028 mol) dissolved in CH_2Cl_2 (50 ml) maintaining the temperature at $-5^{\circ}C$. Stirring was continued at $-5^{\circ}C$ for 5 min and then

105 NaOH (0.34 g, 0.0085 mol) dissolved in water (25 ml) was added under vigorous stirring. The two phases were separated and the aqueous phase was washed with CH₂Cl₂ (10 ml). More CH₂Cl₂ (50 ml) was added to the aqueous phase, the pH was adjusted to 9.5 by

110 adding 2M HCl and after stirring the phases were separated. The organic phase was dried (Na₂SO₄), filtered and the solvent was evaporated off giving an oil which was crystallized from CH₃CN (15 ml) yielding the desired product (0.3 g, 32%), m.p. 161°C.

115 Example 2. Method a. Preparation of 4,6 - dimethyl - 5 - heptyloxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl) methyl] sulfinyl] - 1H - benzimidazole.

m-Chloroperbenzoic acid, 91% (1.13 g, 0.0059 mol)

dissolved in CH_2CI_2 (25 ml) and cooled to $-10^{\circ}C$ was added under stirring to 4,6 - dimethyl - 5 - heptyloxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl) methyl] thio] - 1*H* - benzimidazole (2.7 g, 0.0059 mol) dissolved in CH_2CI_2 (50 ml) maintaining the temperature at $-5^{\circ}C$. Stirring was continued at $-5^{\circ}C$ for 10

125 min. The two phases were separated and then NaOH (0.26 g, 0.0066 mol) dissolved in water (50 ml) was added under vigorous stirring. The two phases were separated. The organic phase was dried (Na₂SO₄), filtered and the solvent evaporated off giving a

130 residual oil, which according to NMR included 30% of

unreacted starting material. The oil was chromatographed on a silica column using CH_3OH — CH_2CI_2 5:95 as eluant and then the product was recrystallized from CH_3CN giving the desired product in crystalline 5 form (0.85 g, 32%), m.p. 116°C.

Which one of these two procedures that have been used for the preparation of the different sulfoxides have been indicated in Table 2 below. For most of the compounds synthesized according to example 2 the 10 chromatographic separation was not performed.

Example 3. Method b. Preparation of 4,6 - dimethyl - 5 - methoxy - 2 - [[(3,4 - dimethyl - 2 - pyridinyl) methyl] thio] - 1H - benzimíďazole.

To 4,6 - dimethyl - 5 - methoxy - 2 - mercapto - 1H15 benzimidazole (1.04 g, 0.0050 mol) in methanol (50 ml) were added (in the following order) NaOH (0.2 g, ;.0050 mol) dissolved in water (2 ml) and 3,4 - dimethyl - 2 - chloromethylpyridine hydrochloride (0.96 g, 0.0050 mol). The mixture was heated until 20 reflux. NaOH (0.2 g, 0.0050 mol) dissolved in water (2 ml) was added dropwise and then the reflux was continued for 3 hours. The mixture was poured on ice-water (200 ml). Filtration and recrystallization from CH₃CN gave the desired product (1.1 g, 67%).
25 NMR data for the final product is given below.

Example 4 and 5. Method d. Preparation of N¹-benzoyl - 5 - methoxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl) methyl] - thio] - 1H-benzimidazole and N¹ - benzoyl - 6 - methoxy - 2 - [[(4 - 30 methoxy - 3,5 - dimethyl - 2 - pyridinyl) methyl] thio] - 1H - benzimidazole

5-Methoxy-2-[[(4-methoxy-3,5-dimethyl-2-pyridinyl) methyl]-thio]-1H-benzimidazole (3.0 g, 0.009 mol) was dissolved in CH₃CN (30 ml) and 35 triethylamine (1.9 ml) was added. Benzoyl chloride (1.4 g, 0.010 mol) was added dropwise under stirring during 15 min. Then the mixture was stirred at 55°C for 45 min. The solvent was evaporated off and ether was added to the residue underice-cooling. The

40 crystalline residue, thus obtained was stirred with water, filtered off and dried giving a white crystalline product mixture (1.9 g, 48%) of the desired two products in a 75:25 molar ratio (according to HPLCanalysis and NMR). NMR data for the final products is 45 given below. Example 6. Method d. Preparation of N - methoxy-carbonyl - 5,6 - methylenedioxy - 2 - [{(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl) - methyl] sulfinyl] - 1H - benzimidazole.

50 Chloro methylformate (0.24 g, 0.0026 mol) dissolved in CH₂Cl₂ (5 ml) was added dropwise to a stirred solution of 5,6 - methylenedioxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl] - methyl] sulfinyl] - 1H - benzimidazole (0.80 g, 0.0022 mol) and triethylamine in CH₂Cl₂ (10 ml). The mixture was then stirred at room temperature for 19 h. The CH₂Cl₂-solution was washed with water, dried (MgSO₄) and the solvent was evaporated giving the desired product as an oil (0.06 g, 6%). NMR data for the final product is given below.

Example 7. Method d. Preparation of N^3 - (N' - phenylcarbamoyl) - 5,6 - methylenedioxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl) - methyl] sulfinyl] - 1H - benzimidazole.

Phenylisocyanate (0.20 g, 0.00167 mol) dissolved in CH₂Cl₂ (5 ml) was added dropwise under stirring to a solution of 5,6 - methylenedioxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl) - methyl] sulfinyl] - 1H-benzimidazole (0.50 g, 0.00139 mol) and triethylamine (0.28 g, 0.00278 mol) in CH₂CL₂ (15 ml). The mixture was then stirred at room temperature for 50 hours. The CH₂Cl₂-solution was washed with water, dried (MgSO₄) and the solvent was evaporated giving the desired product as an oil (0.03 g, 5%). NMR data
for the final products is given below.

Example 8. Method e. Preparation of 4,6-dimethyl - 5 - methoxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl)methyl]sulfinyl] - 1H - benzimidazole.

N¹ - Propionyl - 4,6 - dimethyl - 5 - methoxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl)methyl]sulfinyl] - 1H - benzimidazole (1.0 g, 0.0023 mol) was heated in 1M NaOH (15 ml) for 1 h under stirring and N₂-atmosphere, pH was adjusted to 9.5 by addition of 2M HCl. Extraction with CH₂Cl₂, separation of the phases, drying the organic phase, evaporation of the solvent and recrystallization from CH₃CN gave the

The following Table 2 gives data for further examples of compounds of the invention.

desired product (0.30 g, 35%), m.p. 137°C.

Table Z. Summary of working examples.

Ex	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	Method (Ex. No.)	Yield X	M.p.(^O C) other data
9	S	H	CH3	СН3 .	сн3	CH ₃	н	CH3	OCH2CH=CH2	CH ₃	b (Ex 3)	82	164-165
10	SO	н	СНЗ	СНЗ	• сн ₃	CH ³	н	CH3	och ₂ ch≠ch ₂	CH ₃	a (Ex 2)	73	146-148
11	٤.	H	СНЗ	CH3	CH ₃	CH ₃	н	CH ₃	осн ₃	CH ³	b (Ex 3)	79	207
12	so	h	CH ₃	сн3	сн ₃	СН3	н	СНЗ	осн ₃	СНЭ	a (Ex 2)	32	193
ĘĴ	s	H	CH3	Сн3	CH3	H	н	CH ₃	ссн ₂ сн=сн ₂	снз	b (Ex 3)	97	165
14	SO	H	СНЗ	CH3	CH ₃	н	н	СНЗ	OCH2CH*CH2	CH3	a (Ex 2)	59	147
15	S	H	CH3	СНЗ	CH ₃	н	н	CH3	осн ₃ .	CH3	b (Ex 3)	79	159
16	ŞQ	ħ	СНЗ	CH ₃	CH ₃	н	н	CH3	осн3	сж ³	a (Ex 1)	83	138
17	2	н	СНЗ	CH ₃	н	СНЗ	н	CH ₃	осн ₂ сн=сн ₂	CH ₃	b (Ex 3)	77	NMR

Ex	X	R ¹⁵	R ¹	K2	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ³	Method (Ex. No.)	Yield	M.p.(^O C) other data
18	\$0	н	CH3	CH ³	н	CH ₃	н	Снз	осн ² сн=сн ²	CH3	a (Ex 1)	- 53	129
19	S	н	CH ₃	CH3	h	CH ₃	н	CH3	осн ³	CH ₃	b (Ex 3)	79	163
20	SO	H	CH ₃	CH3	н	CH3	н	CH3	осн3	CH ₃	a (Ex 1)	52	191
21	S	Н	CH ₃	CH3	н	H	H	CH3	OCH2CH=CH2	СНЗ	b (Ex 3)	37	103.
22	SO	Н	CH ³	CH3	н	H	н	CH3	OCH2CH=CH2	CH ₃	a (Ex 1)	58	149
23	S	H	н	СнЗ	CH ₃	H	H	CH3	och ₂ ch-ch ₂	CH ₃	b (Ex 3)	99	181
24	\$0	н	н	CH ³	CH ₃	н	Н	CH3	OCH ² CH=CH ²	СН3	a (Ex 1)	71	157
25	S	H	CH3	H	н	CH ₃	Н	CH3	och ₂ ch∗ch ₂	CH3	b (Ex 3)	62	MR
26	SO	H	сн3	H	ж	CH3	н	CH3	och ₂ ch=ch ₂	сн ₃	a (Ex 1)	10	155
27	5	H	CH ³	H	Н	н	H	CH3	OCH2CH=CH2	СН3 _	b (Ex 3)	90	KAR
28	50	H	CH3	н	н	н	H	CH3	осн ₂ сн=сн ₂	CH ₃	a (Ex 1).	69	142
29	\$	H	H	CH ₃	н	H	н .	CH3	OCH2CH=CH2	CH3	b (Ex 3)	74	NMR
30	S0	H	Н	CH3	н	н	H	CH3	och ₂ ch∗ch ₂	CH3	a (Ex 1)	55	134
31	S	H	н	OCH 3	н	н	н	CH ₃	OCH ² CH=CH ²	CH ₃	b (Ex 3)	51	105-107
12	SO	H	Н	OCH3	н	н	H	CH3	OCH2CH=CH2	CH ₃	a (Ex 1)	62	111
13	\$	H	н	OCH ₃	н	н	H	CH3	OCH ₂ C≇CH	CH ₃	b (Ex 3)	66	154

cent.

Ex	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	Method (Ex. No.)	Yield Z	M.p.(^O C) other data
34	\$0	H	н	осн ₃	н	H	H	CH3	OCH ^S C=CH	us	a (Ex 1)	71	145
35	50	н	Н	CCH3	н	H	н	н	OCH3	C ₂ H ₅	a (Ex 1)	31	147
35	s	H	н	осн ₃	н	H	н	н		-(CH ₂) ₄ -	b (Ex 3)	61	50-IR
37	SO	н	н	осн ³	H	н	н	н		-(CH ₂) ₄ -	a (Ex 2)	34	NHR
38	s	н	н	CH O	, н	н	н	сн ₃	осн3	CH3	b (Ex 3)	22	148
40	s	н	CH3	H	сн ₃	н	н	CH3	OCH ₂ CH=CH ₂	CH ₃	b (Ex 3)	76	134-136
41	50	н	CH3	Ħ	СH ₃	н	н	сн3	OCH2CH=CH2	CH ₃	a (Ex 1)	35	111
42	s	н	H	осн ₂ сн	н	н	н	СНЗ	осн ₃	CH3	b (Ex 3)	29	66
43	SO	H	н	OCH ₂ CH	н	H	н	CH ₃	осн3	CH3	a (Ex 1)	39	. 94
44	\$	н	н	-(○.	н	н	н	сн3	OCH3	CH3	b (Ex 3)	75	RMR
45	50	н	н	\leftarrow	н	н	н	CH ₃	оснз	сн ₃	a (Ex 2)	60	155

Éx	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	₹ 5	R ⁷	R ⁸	Method (Ex. No.)	Yield	M.p.(GC) other cata
47	so	н	н	CEOCH 2 CH 2 OCH 3	CH3	H	н	снз	осн3	CH3	à		
48	s	н	н	COOCH 2	CH3	H	н	CH3	осн ₃	CH3	c		
49	\$0	н	H	COOCH	CH3	н	н	CH ₃	осн ₃	CH3	à		•
50	s	н	н	CH ₂ OH	CH3	н	н	СНЗ	ссн3	CH3	b (Ex 3)	86	192
51	50	н	rt	си2он	CH3	H	н	CH3	ССНЗ	CH ₂	a (Ex 1)	10	169
52	S	н	n 70	CH ² 0CO-	CH3	н	н	CH3	осн ₃	CH ₃	c		
53	50	н	н	CH ² 0CO-	CH ₃	н	н	CH3	осн ₃	CH3	a	•	
54	5	H	H	соосн3	· CH3	н	H	CH ₃	och ₂ ch≠ch ₂	CH3	b (Ex 3)	75	168
55	SO	H	H	соосн3	CH3	Н	H	CH ₃	OCH2CH=CH2	CH.	a (Ex 1)	52	139
56	S	H	CH3	осн ₃	CH ³	н	н	CH3	OCH ³	CH3	b (Ex 3)	70	NMR
8	so	H	CH3	осн ₃	CH3	н	н	CH3	осн ₃	CH3	(a (Ex 3)	56 35	137 137
3	S	H	CH3	осн ₃	CH3	н	H	CH3	CH ₃	н	b (Ex 3)	55 67	NMR
1	SO	н	СНЗ	осн ₃	CH ₃	н	н	CH3	CH ₃	н	a (Ex 1)	32	161
7	S	н	Ch ₃	OCH2CH2OCH3	CH3	н	н	CH3	осн ₃	CH3	b (Ex 3)	90	NMR
8	50	H	сн3	осн ₂ сн ₂ осн ₃	Сн3	н	н	CH3	OCH3	CH3	a (Ex 1)	68	144

Ex	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	Method (Ex. No.)	Yield	M.p.(^O C) other data
59	s	н	снз	осн ₂ сн ₂ осн ₃	CH3	н	н	н	сн3	СНЗ	b (Ex 3)	95	RMR
60	SQ	H	CH ₃	осн ₂ сн ₂ осн ₃	CH ₃	н	н	н	сн3	CH ₃	a (Ex 1)	58	131
61	5	н	CH ₃	сосн3	CH ₃	H	H	СНЗ	осн ₃	CH ₃	b (Ex 3)	90 .	192-4
62	50	H	CH3	COCH ³	снз	н	н	CH ₃	осн ₃	CH3	a (Ex 2)	25	164-5
63	S	H	CH3	сосн3	CH ₃	H	н	CH3	H.	CH ₃	b (Ex 3)	99	184-6
64	SO	н	CH3	сосн	CH3	н	н	СНЗ	н	CH3	a (Lx 2)	91	148-50
65	S	н	CH ₃	COC ₂ H ₅	CH ₃	н	н	СНЗ	OCH 3	CH3	b (Ex 3)	68	. 149
66	50	н	CH3	COC ₂ H ₅	CH3	н	н	CH ₃	осн ₃	CH ₃	a (Ex 2)	48	NMR
67	S	H	CH3	^C 2 ^H 5	CH3	H	н	CH ₃	осн3	CH3	b (Ex 3)	91	182
68	SO	H	сн3.	с ₂ н ₅	CH ₃	H	н	CH3	OCH3	CH ₃	a (Ex 2)	67	175-7
69	S	H	СНЗ	C ₂ ri ₅	CH ₃	н	н	CH ₃	осн ₃	н	b (Ex 3)	95	NHR.
70	SO	H	CH3	C2H5	CH3	H	H	СНЗ	осн ³	н	a (Ex 2)	73	142-3
71	S	H	C2H2	CN	C ₂ H _S	Н	н	CH ₃	осн ₃	CH3	b (Ex 3)	82	150
72	SO	H	c ₂ H _S	Ctt	czHs	H	н	СНЗ	осн ₃	CH ₃	a (Ex 2)	81	180
73	S	H	CH3	OCH3	CH3	CH ₃	н	CH3	осн3	Сн3	b (Ex 3)	82	143
74	50	н	CH ₃	осн3	CH3	СНЗ	H	CH ₃	осн3	СНЗ	a (Ex 2)	43	163

Łх	X	н ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	a ⁷	R ³	Method (Ex. No.)	Yield 2	M.p.(^O C) other data
75	s	н	c1 (CI	Cì	н	н	CH3	осн ₃	СНЗ	b (Ex 3)	90 -	204
76	so	н	cı i	cı • .	CI	н	н	CH3	осн ₃	C:43	a		
77	50	н		снз	СНЗ	н	н	н	ссн3	C2H2	a (Ex 1)	43	156
78	s	н	н (н	н	н	Сн3	осн3	сн3	b (Ex 3)	.90	EMB .
79	so	н		CO:1	н	н	н	сн3	осн ₃	CH3	a (Ex 1)	61	6MR
80	s	н	н	-ссн ₂ о-		н	н	СНЗ	осн ₃	СНЗ	b (Ex 3)	91	168
81	SO	н	н	-осн ₂ о-		н	н	CH ₃	ссн ₃	CH ₃	a (Ex 1)	67	165
82	s	н	-CH=CH-C	н=СН-	н	H	н	СНЗ	CCH3	CH ₃	b (Ex 3)	73	HAS.
83	so	H	-сн=сн-с	H=CH-	H	H	н	СНЗ	och3	CH3	a (Ex 1)	60	184
84	s	ж	н	-CH=CH-CH=CH-		н	н	CH3	осн ₃	CH ₃	b (Ex 3)	78	191
85	SQ	н	н	-сн-сн-сн-сн-		н	н	СНЗ	0CH ₃	CH3	a (Ex 1)	34	175
85	5	н	-сн ₂ сн ₂ с	H ₂ CH ₂ -	н	н	н	CH ₃	осн ₃	CH ₃	b (Ex 3)	58	NMR
87	50	н	-CH2CH2C		н	н	н	CH ₃	осн ₃	CH ₃	a (Ex 1)	27	175
88	s	н	н	-осн ₂ о-		н	CO ₂ CH ₃	CH ₃	OCH ³	СНЗ	đ		

M.p.(^OC) other date R⁷ R8 R15 R1 R² Yield Ex X Method (Ex. No.) OCH3 d (Ex 6) NMR -осн₂о-CO2CH3 CH3 CH3 6 SO H н соин-О сн₃ -0CH₂0d (Ex 7) 5 HMR OCH₃ CH3 7 SO H 0CH2CH2CH2O-€ CH₃ осн3 CH₃ b (Ex 3) 25 MMR осн₂сн₂сн₂о-О 0CH₃ a (Ex 2) 78 61 CH3 CH3 0(CH₂)6CH₃ b (Ex 3) CH₃ сн₃ н CH₃ CCH3 СНЗ 64 . MA CH₃ a (Ex 2) 32 116 0(CH₂)6CH₃ CH₃ CH3 OCH2CH=CH2 45 MAR CH₃ CH₃ b (Ex 3) C2H5 OCH2CH=CH2 124-6 CH3 대3 a (Ex 1) 49 94 SO C_2H_5 осн₂сн₂сн(сн₃)₂ CH3 b (Ex 3) 25 HMR OCH₃ CH3 ссн₂сн₂сн(сн₃)₂ OCH₃ CH3 CH₃ a (Ex 1) 33 111 -сн-сн-сн-с--cн₂сн₂-0CH3 CH3 CH₃ b (Ex 3) 96 190 -сн=сн-сн=С-сн₂сн₂a (Ex 2) 93 109 CH3 OCH3 CH3 0CH₃ d (Ex 4) 4 S CH₃ 0CH₃ CH_3 MAR 5 S н осн₃ н ∞-(○) СНЗ осн3 CH3 d (Ex 5) н

27

Ex.	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	Method (Ex. No.)	Yield	M.p.(^O C) other data
99	s	H	н	CH(CH3)2	н	H	н	СНЗ	осн ⁵ сн-сн ⁵	CH3	b (Ex 3)	99	70
101	\$	н	н	C(CH3)3	н	н	H	СНЗ	осн ₂ сн=сн ₂	CH ₃	b (Ex 3)	52	88-89
102	\$0	ĸ	H	с(сн ₃)3	н	н	н	сн3	осн ₂ сн=сн ₂	CH3	a (Ex 2)	12	TIMR
103	\$	H	н	сн ⁵ сн ⁵ осн ³	н	H	н	сн3	осн3	CH3	b (Ex 3)	84	NMR
104	\$0	н	н	CH2CH2OCH3	_ #	H	H	СНЗ	осн ₃	CH3	a (Ex 1)	38	118
105	s	н	н	->>	جاً_	н	н	сн3	OCH3	Сн3	b (Ex 3)	58	216
106	50	н	н	5	$\tilde{\mathcal{L}}$	н	н	СНЗ	осн3	сн ₃	a (Ex 2)	32	158
107	\$0	В	н	OCH ₃	-0- н	н	CO ² CH ³	СНЗ	осн3	CH ₃	d (Ex 4 and 5	, \{6	\\ NMR
108	SO.	н	Н	н	OCH3	н,	co ₂ cH ₃	CH3	осн ₃	CH ₃	a Con y and s	′ ′ ′	}"""
109	5	н	H	SCH ₃	H	H .	н	CH3	OCH3	. CH ₃	6 (Ex 3)	83	147-148
110	S	н	н	сн(сн ³) ⁵	н	н	H	оч3	OCH Z	CH ³	b (Ex 3)	86	TH NMR
111	SO	н	н	сн(сн ₃)2	н	H	н	СНЗ	OCH 2	снз	a (Ex 2)	89	I _{H NMR}

cont.

Table 2 cont.

K12	R¹	R ²	R³	R ⁴	R ⁵	₽ _P	R ⁷	R ⁸	Method (Ex. No.)	Yield Z	H.p.("C) other data
н	н	сн2сн2сосн3	н	н	н	СНЗ	OCH2CH=CH2	CH3	b (Ex 3)	40	1 _{H NMR}
н	н	сн ₂ сн ₂ сосн ₃	н	н	н	Сн3	och ² ch-ch ³	CH3	a (Ex 2)	28	123-4
H	н	دڙ"	н	Н	н	CH3	осн3	сн ₃	b (Ex 3)	21	162
н	н	осн ₃	н	H	H	-CH=0	н-сн•сн-	н	b (Ex 3)	67	105
н	H	^{осн} 3	н	н	н	-CH=C	:н-сн=сн-	н	a (Ex 1)	66	100
н	н	0-(н	н	н	CH3	осн3	сн ₃	p (Ex 3)	98	122
н	н	∘ ≺⊘	н	н	H	CH3	осн3	СН3	a (Ex2)	80	118
н	н	OCH ₂ CH ₂	н	н	н	сн3	осн3	сн3	b (Ex 3)	80	1 _{H NHR}
н	н	CCH2CH2	Н	H	н	сн ³	осн3	сн ³	a (Ex 2)	55	145 d
н	н	co- (()	н	H	H	сн ₃	OCH3	CH3	b (Ex 3)	82	H NMR
н	н	co -	H	н	H	CH ₃	осн3	СНЗ	a (Ex 2)	24	1H NHR
н	н	$\overline{\bigcirc}$	н	н	н	сн3	осн3	сн ₃	b (Ex 3)	88	158
	H H H H H	H H H H H H H H H H H H H	H H CH ₂ CH ₂ COCH ₃ H H CH ₂ CH ₂ CCCCH ₃ H H CCH ₃ H H OCH ₃ H H OCH ₃ H H OCH ₃ H H OCH ₂ CH ₂ H H CCH ₂ CH ₂ H H CCH ₂ CH ₂ H H CO— H H CO— H H CO—	H H CH ₂ CH ₂ COCH ₃ H H H CCH ₂ CH ₂ COCH ₃ H H H CCH ₃ H H H OCH ₃ H H H OCH ₃ H H H OCH ₂ CH ₂ COCH ₂ H H H CCH ₂ CH ₂ H H H CCH ₂ CH ₂ H	H H CH ₂ CH ₂ COCH ₃ H H H CH ₂ CH ₂ COCH ₃ H H H C C N H H C N H H OCH ₃ H H H H OCH ₃ H H H H OCH ₂ CH ₂ H H H H C N H H C N H H C N H H C N H H C N H H C N H H C N H H C N H H H C N H H H C N H H C N H H H C N H H H C N H H H C N H H H C N H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H H C N H H C N H H H H C N H H H C N H H H C N H H H C N H H H C N H H H N H H C N H H H N H H C N H H H N H H N C N H H H H N H H N C N H H H N H H N C N H H H H N H H N C N H H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N C N H H H N H H N H H N C N H H H H N H H H N H H N H H N C N H H H H N H H H N H H H N H H N H H N H H N H H N C N H H H H N H H H H N H H N H H H N H H H N H H N H H H N H H N H H N H H H N H H N H H N H H N H H N H H N H H N H H H N H H H N H H H N H H H N H H N H H N H H N H H N H H N H H H N H H H N H H N H H H N H H H N H H H H	H H CH ₂ CH ₂ COCH ₃ H H H H CH ₂ CH ₂ COCH ₃ H H H H C C C C C C C C C C C C C C C C	H H CH2CH2COCH3 H H H CH3 H H CH2CH2COCH3 H H H CH3 H H CSO H H H H CH3 H H OCH3 H H H -CH-CH-CH-CH-CH-CH-CH-CH-CH-CH-CH-CH-CH-	H H CH2CH2COCH3 H H H CH3 OCH2CH-CH2 H H CH2CH2COCH3 H H H CH3 OCH3 H H OCH3 H H H CH3 OCH3 H H OCH2CH2	H H CH2CH2COCH3 H H H CH3 OCH3 CH3 H H CH2CH2COCH3 H H H H CH3 OCH3 H H OCH3 H H OCH3 H H H CH3 OCH3 CH3 H H OCH2CH2COCH3 H H H H CH3 OCH3 CH3 H H OCH2CH2COCH3 H H H H CH3 OCH3 CH3 H H OCH2CH2COCH3 H H H H CH3 OCH3 CH3 H H CO-○ H H H H CH3 OCH3 CH3 H H CO-○ H H H H CH3 OCH3 CH3 H H CO-○ H H H H CH3 OCH3 CH3 CH3	H H CH ₂ CH ₂ COCH ₃ H H H CH ₃ OCH ₂ CH ₂ CH ₂ CH ₂ CC CH ₃ b (Ex 3) H H CH ₂ CH ₂ COCH ₃ H H H CH ₃ OCH ₂ CH ₂ CH ₂ CH ₂ C CH ₃ a (Ex 2) H H CSO H H H H CH ₃ OCH ₃ CH ₃ b (Ex 3) H H OCH ₃ H H H CH ₃ OCH ₃ CH ₄ b (Ex 3) H H OCH ₃ H H H CH ₃ OCH ₃ CH ₃ b (Ex 3) H H OCH ₃ H H H CH ₃ OCH ₃ CH ₃ b (Ex 3) H H OCH ₂ CH ₂ O H H H CH ₃ OCH ₃ CH ₃ a (Ex 2) H H OCH ₂ CH ₂ O H H H CH ₃ OCH ₃ CH ₃ a (Ex 2) H H COO H H H H CH ₃ OCH ₃ CH ₃ a (Ex 2) H H COO H H H H CH ₃ OCH ₃ CH ₃ a (Ex 2)	H H CH ₂ CH ₂ COCH ₃ H H H CH ₃ OCH ₂ CH ₂ CH ₂ CH ₂ CH ₃ b (Ex 3) 40 H H CH ₂ CH ₂ CGCH ₃ H H H CH ₃ OCH ₂ CH ₂ CH ₂ CH ₂ CH ₃ a (Ex 2) 28 H H C H H CH ₃ H H H CH ₃ OCH ₃ CH ₃ b (Ex 3) 21 H H OCH ₃ H H H CH ₃ OCH ₃ CH ₄ b (Ex 3) 67 H H OCH ₃ H H H CH ₃ OCH ₃ CH ₃ b (Ex 3) 98 H H O H H H CH ₃ OCH ₃ CH ₃ CH ₃ b (Ex 3) 98 H H O H H H CH ₃ OCH ₃ CH ₃ CH ₃ b (Ex 3) 80 H H OCH ₂ CH ₂ O H H H CH ₃ OCH ₃ CH ₃ b (Ex 3) 80 H H CO H H H CH ₃ OCH ₃ CH ₃ CH ₃ b (Ex 3) 80 H CH ₃ OCH ₃ CH ₃ CH ₃ b (Ex 3) 82 H H CO H H H CO H H H CH ₃ OCH ₃ CH ₃ CH ₃ b (Ex 3) 82

Table 2 cont.

E.	x	R ¹⁵	R.I	R ²	R ³	R ³	R ⁵	R ⁶	R ⁷	R8	Method Ex. No.)	Yield "	M.p. ("C) other data
124	50	н	н	$\overline{\bigcirc}$	•, н	н	н	СНЗ	осн3	снз	a (Ex 2)	52	104
125	s	н	н	SOCH ₃	н	н	н	СН3	осн ₃	CH ₃	b (Ex 3)	57	1 _{H SMR}
126	SÒ	н	н	SOCH ₃	, н	н	н	сн3	осн ₃	CH ₃	a (Ex 1)	47	H SER
127	90	н	н	NO ₂	н	н	н	СНЗ	OCH3	CH3	a (Ex 1)	14	1 _{H NMR}
128	s	н	н	Br	н	н	н	CH3	OCH ² CH=CH ²	CH3	b (Ex 3)	64	171
129	so	н	н	8r	н	н	н	CH ₃	och ² ch•ch ²	сн ₃	a (Ex 2)	58	143
130	\$	н	н	осн3	н	н	н	-CH+C	:H-0-	н	b (Ex 3)	77	MMR
131	S0	н	н	асн _з	н	н	н	-CH=C	H-0-	н	a (Ex 2)	19	NMR
132	so	н	н	CH ³	сн ₃	н	ÇOC (CH ³)	3 CH3	осн ₃	снз	d (Ex 6)	22	168
134	SO	н	н	сн3	снз	н	си(сн ³) ⁵	СНЗ	осн3	снз	d (Ex 6)	21	¹ H NHR
135	s	н	н	сн ₃	CH3	H	н .	сн3	OCH 2 O	сн3			
136	92	н	H	сн3	снз	11	н	сн3	OCH 2	снз			
137	s	н	н		-сн ² сн ³ сн ³ -	н	н	CII3	осн3	CH3	b (Ex 3)	74	160
138	Sσ	н	11		-CII2CII2CII2-	н	н	СНЗ	ocii3	снз	a (Ex 1)	40	171

. Table 2 cont.

Ex	x	R ¹⁵	R1	R ^Z	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	Method (Ex. No.)	Yield Z	M.p. (°C) other data
139	s	н	-CH=	CH-CH=N-	н	н	н	сн3	осн3	снз	b (Ex 3)	38	EMR
140	50	H	-CH=	CH-CH=N-	н	H	н	CH ₃	осн3	CH3	a (Ex 1)	26	60
141	s	н	н		-осн ₂ о-	н	н	CH ₃	CH3	CH ₃	b (Ex 3)	83	193-95
142	SO	н	н	a	осн ⁵ 0	н	н	CH3	сн3	CH3	a (Ex 2)	76	173
143	SO	н	н	сосн ₃	CH ₃	н	н	н	осн3	с ₂ н ₅	a (Ex 2),	49	154
144	S	н	CH3	CH3	CH3	н	н	CH ₃	сн ₃	н	b (Ex 3)	39	1 _{H NMR}
145	\$0	н	CH3	СНЗ	CH3	н	H	CH3	CH ₃	н .	'a (Ex 2)	65	¹ H RMR
146	s	н	СНЗ	CH ₃	CH3	H	н	н	сн3	CH3	b (Ex 3)	78	143
147	SO	H	СНЗ	сн3	CH3	н	н	H	CH3	сн ₃	a (Ex 2)	64	180
148	s	н	CH3	сн ₃	CH3	н	н	CH3	H	CH ₃	b (Ex 3)	70	239-42
49	SO	H	CH ₃	сн3	CH3	H	н	CH3	н	CH3	a (Ex 2)	14	171
50	s	H	CH3	СНЗ	H	СНЗ	н	CH3	CH3	н	b (Ex 3) .	56	210
151	50	н	СНЗ	сн3	н	CH ₃	н	CH ₃	СНЗ	н	a (Ex 2)	66	1H KMR
152	s	н	CH ₃	CN	CH ₃	н	н	СНЗ	0C2H5	CH3	b (Ex 3)	94	151
153	S0	н	CH3	Cii	сн3	н	н	CH3	^{QC} 2 ^H 5	CH3	1 (Ex 2)	29	150
154	s	н	H	\leftarrow	7′ н	н	H	н	СНЗ	C2H5	b (Ex 3)	48	1 _{H 125} R

Tab!	 •	cont	

Ex X	R 15	R ¹	R ²		R3	R [‡]	_R 5	R ⁶	R ⁷	_R 8	Mentod (Ex. No.)	Yield 2	H.p. ('C) other data
155 50	н	н	-	7.	н	н	н	н	Сн3	C2H5	a (Ex 2)	44	105
156 S	н	н	$\overline{\bigcirc}$		н	H	н	Сн3	осн ₂ сн ₂ осн	сн3	b (Ex 3)	94	1 _{H NMR}
157 SO	н	н	$\overline{}$		H	н,	н	CH3	осн2сн5осн	сн3	a (Ex 2)	18	181
158 S	H	н	CF ₃		н	н	н	сн3	OCH 2	сн3	b (Ex 3)	67	100
159 SO	н	н	CF ₃		н	н	н	СНЗ	OCH ²	сн3	a (Ex 2)	57	125
160 S	н	H	сн ₂ сн ₂ соос ₂ н ₅		H	H	н 0	сн ₃	осн3	сн ₃	b (Ex 3)	15	1 H NMR
161 50	н	н	осн3		н	н	Ç-0C(CH	3 ³ CH ₃	осн ₃	снз	d (Ex 6)	50	155
163 50	н	н	осн3		н	H	H	-сн ₂ с	H ₂ 0-	н			
164i S	н	н	осн ₃		н	н	H	-сн ₂ с	H2CH20-	н	b (Ex 3)	71	1 H NMR
165 50	H	н	OCH ₃		н	н	H	н		-осн ₂ сн ₂ -			
166 SO	H	H	осн3		н	H	н	н		-осн ₂ сн ₂ сн ₂ -			

Identifying data for compounds of the invention

MMR-data of the compounds in Table 2 (90 MHz)

NMR-data: \$\((COC1_3\)\) ppm 2.3(s,3H), 2.35(d,6H), 2.5(s,3H), 2.55(s,3H), 4.4(s,2H), 4.25-4.4(d,2H), 5.2-5.6(m,2H), 5.2-6.4(m,1H), 6.9(s,1H), 8.35(s,1H).
4.4(s,2H), 4.25-4.4(d,2H), 5.2-5.6(m,2H),
•
2.2(s,3H), 2.3(s,3H), 2.6(s,3H), 4.35-4.45(d,2H), 4.45(s,2H), 5.2-5.6(m,2H), 5.85-6.35(m,1H), 6.9-7.55(m,3H), 8.3(s,1H).
2.2(s,3H), 2.25(s,3H), 2.4(s,3H), 4.2-4.35(c,2H), 4.4(s,2H), 5.5-5.6(m,2H), 5.85-6.3(m,1H), 6.9-7.1(d,1H), 7.3-7.55(t,2H), 8.3(s,1H).
1.8(m,4H), 2.75(m,4H), 3.8(s,3H), 4.25(s,2H), 6.95(m,1H), 7.05(s,2H), 7.4(d,1H), 8.3(s,1H).
1.7(m,4H), 2.3-2.7(m,4H), 3.85(s,3H), 4.6(d,2H), 6.8(s,1H), 7.05(s,2H), 7.6(m,1H), 8.3(s,1H).
1.2-2.0(m,10H), 2.25(s.3H), 2.3(s,3H), 2.6(m,1H), 3.75(s,3H), 4.45(s,2H), 7.1(q,1H), 7.5(m,2H), 8.35(s,1H).

MMR-data of the compounds in Table 2. (cont.)

Example	
110 -	NNR-data: &(CDC13) ppm
3	2.3(s,6H), 2.35(s,3H), 2.5(s,3H), 3.75(s,3H),
1	4.4(s,2H), 7.05-7.2(d,1H), 7.25(s,1H),
	8.3-8.45(d,1H).
57	2.2(s,3H), 2.25(s,3H), 2.3(s,2H), 2.5(s,3H),
]	3.45(s,3H), 3.75(s,3H), 3.85(m,4H), 4.3(s,2H),
	7.2(br.s., 1H), 8.3(s,1H).
59	2.3(s,6H), 2.4(s,3H), 2.55(s,3H), 3.5(s,3H),
1	3.9(m,4H), 4.3(s,2H), 7.2(s,1H), 7.3(s,1H),
	8.4(s,1H), 9.3(br.s., 1H).
86	1.2(t,3H), 2.15(s,3H), 2.2(s,3H), 2.3(s,3H),
	2.4(s,3H), 2.8(q,2H), 3.65(s,3H), 4.8(s,2H),
	7.3(s,1H), 8.25(s,1H).
69	1.1(t,3H), 2.2(s,3H), 2.4(s,3H), 2.55(s,3H),
	2.75(q,2H), 3.85(s,3H), 4.35(s,2H), 6.75(d,1H),
	7.25(s,lH), 8.4(d,lH).
78	1.2(d,3H), 1.6(m,6H), 2.25(s,3H), 2.3(s,3H),
	3.0(m,1H), 3.75(s,3H), 4.15(m,1H), 4.45(s,2H),
	4.55(m,1H), 7.3(q,1H), 7.5(m,2H), 8.3(s,1H).
79	1.25(d.3H), 1.65(m,6H), 2.15(s,3H), 2.2(s,3H),
	3.1(m,1H), 3.65(s,3H), 4.1(m,1H), 4.6(m,1H),
	4.8(s,2H), 7.4(q,1H), 7.7(d,1H), 7.8(s,1H),
	8.3(s,1H).
82	2.2(s,3H), 2.3(s,3H), 3.7(s,3H), 4.75(s,2H),
	7.3-8.5(m,8H).

.

IME-cata of the compounds in Table 2. (cont.)

Example	<u> </u>
`. . .	NMA-data: d(COCl ₃) ppm
36	1.85(m,4H), 2.2(s,3H), 2.25(s,3H), 2.7-3.1(m,4H), 3.75(s,3H), 4.35(s,2H), 6.9(d,1H), 7.3(d,1H),
	6.25(s,1H).
6	2-1(s,3H), 2-35(s,3H), 3-8(s,3H), 4-15(s,3H),
	4.75(s,2H), 6.1(s,2H), 7.3(s,1H), 7.5(s,1H), 8.15(s,1H).
7	2.15(s,3H), 2.2(s,3H), 3.7(s,3H), 4.7(s,2H),
	6.05(s,2H), 7.0-7.6(m,7H), 8.15(s,1H), 8.3(s,1H).
90	2.25(s,3H), 2.1-2.4(m,2H), 2.3(s,3H), 3.75(s,3H),
	4.2(t,4H), 4.4(s,2H), 6.75-7.2(m,SH), 7.2-7.5(m,3H), 8.35(s,1H).
92	0.7-2.05(m,13H), 2.25(s,3H), 2.3(s,3H), 2.35(s,3H),
	2.5(s,3H), 3.65-3.9(m,2H), 3.75(s,3H), 4.35(s,2H), 7.2(s,1H), 8.3(s,1H).
93	
"	1.25(t,3H), 2.25(s,3H), 2.3(s,3H), 2.8(q,2H), 4.4(d,2H), 4.45(s,2H), 5.2-5.65(m,2H), 5.85-6.3(m,1H)
	7.0-7.65(m,2H), 7.5(s,1H), 8.35(s,1H).
95	0.9(s,3H), 1.0(s,3H), 1.5-1.95(m,2H), 2.15-2.45(m,1H), 2.25(s,3H), 2.3(s,3H), 3.7-4.0(t,2H), 3.85(s,3H),
	4.45(s,2H), 2.8-7.0(m,1H), 7.15(d,1H), 7.45-7.55 (d,1H), 8.3(s,1H).
4 - 5	2.25(s,3H), 2.40(s,3H), 3.6 and 3.85(2s, total 3H).
	3.80(s,3H), 4.8 and 4.85(Zs,total ZH), 6.35-7.95 (m,8H), 8.35(s,1H).

MMR-data of the corpounds in Table 2. (cont.)

Ехапріе	
No.	NMR-data: c(CDC)3) ppm
103	2.3(s,3H), 2.35(s,3H), 3.0(t,2H), 3.35(s,3H), 3.65(t,2H), 3.8(s,3H), 4.4(s,2H), 6.8-7.6(m,4H), 8.25(s,1H).
107+108	2.2(s,3H), 2.35(s,3H), 3.75(s,3H), 3.9 and 3.95 {2s,total 3H), 4.15(s,3H), 4.75(s,2H), 7.07-7.95 {m,3H}, 8.15(s,1H).
102	1.32(s,9H), 2.08(s,3H), 2.15(s,3H), 4.09(d,2H), 4.74(s,2H), 5.10-5.45(m,2H), 5.73-6.25(m,1H), 7.28-7.73(m,3H), 8.27(s,1H).
139	2.22(s,3H), 2.29(s,3H), 3.75(s,3H), 4.40(s,2H), 7.38-7.58(m,1H), 7.87-8.02(m,2H), 8.29-8.47(m,1H), 8.70-9.00(m,2H),
110	1.25(d,6H), 1.6-2.15(m,4H), 2.25(s,3H), 2.3(s,3H), 3.0(m,1H), 3.7-4.05(m,4H), 4.25(m,1H), 4.5(s,2H), 7.15(q,1H), 7.5(s,1H), 7.55(d,1H), 8.3(s,1H).
111	1.3(d,6H), 1.55-2.15(m,4H), 2.2(s,3H), 2.25(s,3H), 3.05(m,1H), 3.65(d,2H), 3.9(m,2H), 4.2(m,1H), 4.8 (s,2H), 7.3(d,1H), 7.4-7.8(m,2H), 8.3(s,1H).
119	2.3(s,3H), 2.35(s,3H), 3.15(t,2H), 3.7(s,3H), 4.25(t,2H), 4.4(s,2H), 6.9(q,1H), 7.15(d,1H), 7.3- 7.6(m,6H), 8.35(s,1H).
125	2.3(s,3H), 2.35(s,3H), 2.8(s,3H), 3.8(s,3H), 4.5 (s,2H), 7.5(d,1H), 7.75(d,1H), 8.05(s,1H), 8.4(s,1H).

HMR-data of the compounds in Table 2. (cont.)

Example	
:io.	NMR-data: d(CDCl ₃) ppm
126	2.2(s,6H), 2.8(s,3H), 3.7(s,3H), 4.85(s,2H), 7.6 (q,1H), 7.85(d,1H), 8.15(s,1H), 8.25(s,1H).
127	2.25(d,6H), 3.75(s,3H), 4.9(d,2H), 7.8(d,1H), 8.3(s,1H), 8.3(q,1H), 8.65(d,1H).
134	2.2(d,6H), 2.35(d,6H), 3:1(s,6H), 3.7(s,3H), 4.95 (s,2H), 7.2(s,1H), 7.6(s,1H), 8.3(s,1H).
112	2.1(s,3H), 2.25(s,3H), 2.3(s,3H), 2.65-3.2(m,4H), 4.4(d.2H), 4.42(s,2H), 5.2-5.6(m,2H), 5.9-6.4(m,1H), 7.1(dd,1H), 7.4(d,1H), 7.5(d,1H), 8.35(s,1H).
121	2.25(s,3H), 2.35(s,3H), 3.8(s,3H), 4.45(s,2H), 7.45-8.0(m,7H), 8.15(s,1H), 8.4(s,1H).
122	2.2(s,6H), 3.7(s,3H), 4.8(d,2H), 7.5-8.05(m,7H), 8.2(s,1H), 8.25(s,1H).
144	2.25(s,3H), 2.35(s,6H), 2.38(s,3H), 2.55(s,3H), 4.4(s,2H), 7.15(d,1H), 7.3(s,1H), 8.4(d,1H).
145	2.15(s,3H), 2.23(s,3H), 2.27(s,3H), 2.4(s,3H), 2.47(s,3H), 4.8(s,2H), 7.1(d,1H), 7.3(s,1H), 8.37(d,1H).
151	2.2(s,3H), 2.23(s,3H), 2.35(s,3H), 2.4(s,3H), 2.47(s,3H), 4.8(d,2H), 7.0(s,1H), 7.1(d,1H), 8.37 (d,1H).
130	3.85(s,3H), 4.65(s,2H), 6.8-7.8(m,7H), 8.55(d,1H)

NMR-data of the compounds in Table 2. (cont.)

Example No.	NMR-data: ど(CDCl ₃) ppm
131	3.85(s,3H), 4.95(d,2H), 6.65-7.60(m,7H), 8.45(d,1H).
160	1.15(t,3H), 2.20(s,3H), 2.27(s,3H), 2.49-2.73(m,2H), 2.89-3.13(m,2H), 3.72(s,3H), 4.09(q,2H), 4.37(s,2H), 6.98 and 7.08(dd,1H), 7.30-7.55(m,2H), 8.28(s,1H).
154	1.1-2.1(m,13H),2.3(s,3H),2.5-2.8(m,3H), 4.4(s,2H), 7.1-7.65(m,4H), 8.5(s,1H)
156	1.1-2.0(m,11H), 2.25(s,3H), 2.3(s,3H), 3.45(s,3H), 3.7(t,2H), 4.0(t,2H), 4.4(s,2H), 7.05-7.65(m,3H), 8.35(s,1H)
164 (270 MHz)	2.13(m,2H),2.88(t,2H),3.82(s,3H),4.26(t,2H), 4.69(s,2H),6.7-6.85(m,2H),7.04(d,1H), 7.39(d,1H),8.1(d,1H).

Preparation of intermediates
Example 11. Method A. Preparation of 4,5,7trimethyl - 2 - mercapto - 1H - benzimidazole.

2 - Nitro - 3,4,6 - trimethylaniline (10.2 g, 0.057 mol)

5 was dissolved in 95% ethanol (900 ml) and hydrogenated in the presence of Pd/C-catalyst until the theoretical amount of hydrogen had been consumed (1 hour). The whole mixture was transferred to another flask and potassium ethylxanthate (12.8 g, 0.080 mol) dissolved in water (12.5 ml) was added. The mixture was refluxed overnight, 2M NaOH (20 ml) was added and the volatiles were evaporated off. The residue was dissolved in methanol (300 ml) and the catalyst was filtered off. Part of the solvent (200 ml) was evaporated off. Water (100 ml) was added and the mixture was acidified with acetic acid (10 ml) dissolved in water (20 ml). The crystalline precipitate was filtered off, washed with water and dried under

reduced pressure, giving the desired product (7.2 g, 20 66%), NMR: δ(COCI₃) 2.0(s,3H), 2.05(s,3H), 2.1(s,3H), 3.3(br.s,1H), 6.5(s,1H).

Example 12. Method B. Preparation of 4,6,7 - trimethyl - 5 - methoxy - 2 - mercapto - 1H-

benzimidazole.

A solution of 4 - methoxy - 3,5,6 - trimethyl - 1,2 - phenylenediamine (1.8 g, 0.010 mol) and triethylamine (2.1 g), 0.021 mol) in CHCl₃ (15 ml) was added dropwise to a stirred solution of thiophosgene (0.60 g, 0.0052 mol) in CHCl₃ (5 ml). The mixture was then stirred at room temperature for 1 hour. Water (15 ml) and triethylamine (0.5 g) was added and the mixture was stirred for 1 hour. The precipitate was filtered off, washed with water and dried in the air giving the desired product (0.96 g, 43%), NMR: δ(COCl₃)

35 2.5(s,3H), 2.65(s,6H), 3.65(s,3H), 12.0(br.s.,1H). Example 13. Method C. Preparation of 4-allyloxy - 3,5 - dimethyl - 2 - pyridinyl - methanol.

4 - Allyloxy - 2,3,5 - trimethyl - pyridine N-oxide (4.0 g, 0.021 mol) was added dropwise under stirring to acetic anhydride (8.0 ml, 0.062 mol) preheated to 80°C, giving a final temperature of 120°C. The mixture was then heated at 80°C for 1 hour. Methanol (15.0 ml) was added and the mixture was kept at 80°C for 15 min. The volatiles were evaporated under reduced 45 pressure. 10% HCI (20ml) was added and the mixture was heated at 90°C for 1 hour and then cooled to room temperature. Excess 2M NaOH was added and the mixture was extracted with CH2Cl2. The organic phase was separated out and dried. Volatiles were 50 evaporated off giving the desired product as an oil (3.0 g, 75%), NMR: δ(COCl₃) 2.1(s,3H), 2.25(s,3H), 4.4(m,2H), 4.65(s,2H), 4.75(s, 1H), 5.2-5.65(m,2H), 5.9-6.45(m,1H), 8.3(s,1H).

Example 14. Method D. Preparation of 4 - allyloxy - 3,5 55 - dimethyl - 2 - pyridinyl - methyl chloride hydrochloride.

Thionyl chloride (4.0 ml) dissolved in CH₂Cl₂ (12 ml) was added dropwise to a stirred solution of 4 - allyloxy - 3,5 - dimethyl - 2 - pyridinylmethanol (8.0 g, 0.041 mol) in CH₂Cl₂ (50 ml), maintaining the temperature below 6°C. Then the mixture was stirred at room temperature for 45 min (final temperature 15°C). Isopropanol (2 ml) was added and the solution was heated shortly at 35°C. The solvent was evaporated off and the crystalline residue was recrystallized from ethanol/ether giving the desired product (3.0 g, 29%), m.p. 115°C.

Table 3a. Intermediates. Summary of working examples.

$$R^{2a}$$

$$R^{3a}$$

$$R^{4a}$$

$$R^{5a}$$

$$R^{5a}$$

No.	Z ^{la}	R ^{la}	R ^{2a}	R ^{3a}	R ^{4a}	R ^{5a}	Method ^{x)} (Ex. No.)	Yield (%)	Mp (^O C) other data
IS	SH	сн3	сн ₃	сн3	СНЗ	н	A(Ex I1)	19	BMR
16	SH	CH3	сн3	CH ₃	Н	н	A(Ex 11)	66	RMR
11	SH	СНЗ	сн3	н	СН3	н	A(Ex I1)	66	621R
17	SH	н	~~``	н	н	н	A(Ex 11)	71	NEER
18	SH	CH ₃	OCH3	CH3	н	н	A(Ex I1)	78	RMR
19	SH	CH3	осн ₂ сн ₂ осн ₃	СНЗ	н	н	A(Ex [])	85	RMR
110	SH	CH3	^C 2 ^H 5	сяз	H	H	A(Ex [1)	89	HMR
111	SH	H 8]	CCH2CH2CH2O-	н	н	н	A(Ex 11)	14	167
112	SH	CH3	о(сн ₂)6сн3	CH3	н	н	A(Ex II)	73	MER
! 2	SH	сн3	осн ₃	CH3	СНЗ	н	B(Ex 12)	43	IJMR
113	SH	-C!	н=СH-СH=СH-СH ₂ CH ₂ -		н	н	A(Ex I1)	23	tima

^{*)}Method A: The 1,2-phenylenediamine is reacted with C₂H₅OCS₂K Method B: The 1,2-phenylenediamine is reacted with CSC1₂

No.	Z ^{Za}	д ^{6а}	R ^{7a}	R ^{8a}	Salt/Base	Method XX) (Ex. No.)	Yield (%)	Mp (^O C) other data
[3	ОН	сн3	осн ₂ сн=сн ₂	СНЗ	Base	C(Ex 13)	75	HMR
14	Cl	CH3	осн ₂ сн•сн ₂	СНЗ	HC1	0(Ex 14)	29	1150
114	ОН	CH3	осн ₂ с≡сн	сн3	Base	C(Ex 13)	88	70 ⁰
115	C1	снз	осн ₂ с≡сн	CH ₃	HC1	D(Ex 14)	76	135°
116	ОН	Н	-(CH ₂) ₄	•	Rase	C(Ex 13)	35	NMR
117	C1	н	-(CH ₂)4	-	нст	D(Ex 14)	72	AMR
118	OH	СН3	осн ₂ сн ₂ сн(сн ₃)	, CH ₃	Base	C(Ex 13)	51	IEER
119	Cl	сн3	осн ₂ сн ₂ сн(сн ₃)	CH ₃	HC1	0(Ex 14)	95	
120	OH	сн ₃ (осн ₂ (о)	. сн ₃	Base	C(Ex 13)	30	tens
:21	Cl	сн _з с	CH _Z	.CH3	HC1	9(Ex 14)	23	133
:22	ΗÛ	сн ₃ с	15 ² H ²	CH ³	Base	C(Ex 13)	70	В.р. 120- 26°C/G.4 mm
123	Cl	СН3 С	C2H2	СНЗ	HC1	0(Ex 14)	89	157
124	ОН	-CH=C	H-0-	н	8ase	C(Ex [3)	18	1H NMR
125	C1	-CH=C	H-0-	н	HC1	D(Ex [4)	95	195

xx) Method C: Rearrangement of the pyridine N-oxide with (CH₃CO)₂O. Method O: Chlorination with SOCI₂.

35

55

NMR—data of the compounds in Table 3a and Table 3b
Example

No. NMR-data: δ(ppm) 5 15 δ(DMSO-d₆) 2.05(s,6H), 2.2(s,6H).

16 δ(CDCl₃) 2.05(s,3H), 2.15(s,3H), 2.2(s,3H, 3.2(s,2H), 6.7(s,1H).

δ(CDCl₃) 2.0(s,3H), 2.05(s,3H), 2.1(s,3H),3.3(br.s.,1H), 6.5(s,1H).

10 17 δ(DMSO-d₈) 1.1-2.05(m,10H), 2.4(m,1H), 6.85-7.05(m,3H).

18 δ(DMSO-d₆) 1.95(s,3H), 2.0(s,3H), 3.35(s,3H), 6.55(s,1H).

19 δ(CDCl₃) 2.1(s,3H), 2.15(s,3H), 3.2(s,3H), 3.35-3.8(m,4H), 6.6(s,1H).

110 δ(CDCl₃+DMSO-d₆) 1.05(t,3H), 2.3(s,3H), 2.35(s,3H),

2.6(q,2H), 6.85(s,1H).

5(CDCl₃) 0.5-1.7(m,13H), 2.0(s,3H), 2.1(s,3H),
 3.15(s,2H), 3.35-3.6(m,2H), 6.6(s,1H).

12 δ(CDCl₃) 2.5(s,3H), 2.65(s,6H), 3.65(s,3H), 12.0(br.s.,1H).

113 δ(CDCl₃) 3.35(s,2H), 3.4(s,2H), 7.15-8.05(m,4H), 12.65(br.s.,1H), 13.3(br.s.,1H).

25 13 δ(CDCl₃) 2.1(s,3H), 2.25(s,3H), 4.4(m,2H),
 4.65(s,2H), 4.75(s,1H), 5.2-5.65(m,2H),
 5.9-6.45(m,1H), 8.3(s,1H).

116 δ(CDCl₃) 1.5-1.9(m,4H), 2.5-2.8(m,4H), 4.7(s,2H), 7.3(s,1H), 8.2(s,1H).

30 117

15

118 δ(CDCl₃) 1.0(s,3H), 1.05(s,3H), 1.5-2.05(m,3H), 2.15(s,3H), 2.3(s,3H), 3.75-4.0(t,2H),

4.15-4.5(br.s.,1H), 4.65(s,2H), 8.3(s,1H). 120 δ(CDCl₃) 1.7-2.2(m,4H), 2.15(s,3H), 2.25(s,3H),

3.75- 4.05(m,4H), 4.15-4.4(m,1H), 4.6(s,2H), 8.25(s,1H).

124 δ(CDCl₃) 8.55(d,1H), 7.8(d,1H), 7.5(d,1H), 7.0(d,1H), 5.1(s,2H).

Pharmaceutical preparations containing a compound 40 of the invention as active ingredient are illustrated in the following examples.

Example 167. Syrup

A syrup containing 1% (weight per volume) of active substance was prepared from the following

45 ingredients: 4,6-Dimethyl-5-ethyl-2-[((4-methoxy-3,5-dimethyl-2-pyridinyl)methyllthiol

	3,5-anneuryt-2-pyriamyt/metnyt/tntoj-	
	1H-benzimidazole-HCl	1.0 g
	Sugar, powder	30.0 g
50	Saccharine	0.6 g
	Glycerol	5.0 g
	Flavouring agent	0.05g
	Ethanol 96%	5.0 g
	Distilled water g.s. to a final volume of	100 ml

Sugar and saccharine were dissolved in 60 g of warm water. After cooling the acid addition salt was dissolved in the sugar solution and glycerol and a solution of flavouring agents dissolved in ethanol ware added. The mixture was diluted with materials

60 were added. The mixture was diluted with water to a final volume of 100 ml.

The above given active substance may be replaced with other pharmaceutically acceptable acid addition salts.

10

15

45

Example 168. Enteric-coated tablets

An enteric-coated tablet containing 20 mg of active compound was prepared from the following ingredients: I 5,6-Methylenedioxy-2-[[(4-methoxy-

3,5-dimethyl-2-pyridinyl)methyl]sulfinyl]-200 g 1*H*-benzimidazole 700 g Lactose 6g Methyl cellulose Polyvinylpyrrolidone cross-linked 50 g 15 g Magnesium stearate Sodium carbonate 6g Distilled water a.s. 200 g II Cellulose acetate phthalate 15 g Cetyl alcohol 2000 g Isopropanol Methylene chloride 2000 g

5,6 - Methylenedioxy - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl)methyl]sulfinyl] - 1H - benzimidazole, powder, was mixed with lactose and granulated with a water solution of methyl cellulose and sodium carbonate. The wet mass was forced through a sieve and the granulate dried in an oven. After drying the granulate was mixed with polyvinyl-pyrrolidone and magnesium stearate. The dry mixture was pressed into tabled cores (10 000 tablets), each tablet containing 20 mg of active substance, in a tabletting machine using 6 mm diameter punches.

II A solution of cellulose acetate phthalate and cetyl alcohol in isopropanol/methylene chloride was sprayed onto the tablets I in an Accela Cota, Manesty (RTM) coating equipment. A final tablet weight of 110 mg was obtained.

Example 169. Solution for intravenous administra-35 tion

A parenteral formulation for intravenous use, containing 4 mg of active compound per ml, was prepared from the following ingredients:
4,6-Dimethyl-5-ethyl-2-[[(4-methoxy40 3,5-dimethyl-2-pyridinyl)methyl]thio]1H-benzimidazole 4g
Polyethylene glycol 400 for injection 400 g
Disodium hydrogen phosphate q.s.
Sterile water to a final volume of 1000 ml

4,6 - Dimethyl - 5 - ethyl - 2 - [[(4 - methoxy - 3,5 - dimethyl - 2 - pyridinyl)methyl]thio] - 1H - benzimidazole was dissolved in polyethylene glycol 400 and 550 ml of water was added. pH of the solution was brought to pH 7.4 by adding a water solution of disodium hydrogen phosphate and water was added to a final volume of 1000 ml. The solution was filtered through a 0.22 μm filter and immediately dispensed into 10 ml sterile ampoules. The ampoules were

Biological tests

I. Inhibiting effect in vitro on acid secretion in isolated rabbit gastric glands

Test Method

60 Gastric gland preparation

Isolated rabbit gastric glands were prepared as described by Berglindh et al., Acta physiol. scand. 1976. 96. 150-159. This method involves vascular perfusion of the rabbit stomach via the gastric arteries, scraping and scissor mincing of the sepa-

rated gastric mucosa and collagenase (0.1%, Type I, Sigma Chemicals, St. Louis, MO. USA) digestion at 37°C for 60-90 min. The glands are then harvested and filtered through nylon cloth to remove coarse fragments. The glands are thereafter incubated at 37°C in a medium containing NaCl 132.4 mM, KCl 5.4 mM, NaH₂PO₄, 5.0 mM, NaH₂PO₄, 1.0 mM, MgSO₄ 1.2 mM, CaCl₂ 1.0 mM, glucose 10 mM, and 1 mg/ml rabbit albumine, pH 7.4.

75 Measurement of acid secretion

The acid secretion in the isolated gland preparation was recorded by measuring the uptake of ¹⁴C-labelled aminopyrine into the glands as described by Berglindh et al., Acta physiol. scand. 1976. 97. 401-414. Accumulation of aminopyrine in the glands indicates gastric acid secretion within the glands. The standard medium contained 10-6M 14C-aminopyrine (Amersham, Great Britain). After the incubation period, the glands were centrifuged, the supernatant was removed and the glands dried, weighed and dissolved in Soluene -350 (Packard, IU. USA). Samples of the supernatant and glands were separately counted in a scintillation counter. The accumulation of 14C-labelled aminopyrine in the glands was calculated as detailed by Berglindh et al., Acta physiol. scand. 1976. 97.403.

Experimental protocol

Glands were incubated for 60 min. in the presence of 5 × 10⁻⁵M histamine and the test compound to be studied. The free base of the test compound was dissolved in methanol. The final concentration of methanol was 1% in the incubation medium, having no influence on the aminopyrine accumulation ratio. For each test compound a complete dose-response curve was generated by testing doses in duplicate in the concentration range 10⁻⁷M to 10⁻⁴M. The logarithm of the concentration (in M) of the test compounds giving 50% inhibition of the aminopyrine accumulation in the glands (IC₅₀) is listed in Table 4 below.

II. Inhibiting effect in vivo on gastric acid secretion in conscious dog

Test Method

Chronic gastric fistula dogs were used. These dogs
110 have been surgically provided with a gastric cannula
in the stomach and a duodenal fistula used for direct
introduodenal administration of test compounds.
Following a 4 weeks' recovery period after surgery,
tests were performed once a week on each dog. Food
115 and water were withdrawn 18 hours before each test.

Gastric acid secretion was induced by continuous infusion of histamine at individual doses (100-300 nmol/kg, h), resulting in submaximal secretion of gastric acid. At least 2 hours after onset of stimulation, when the gastric acid secretion had reached a steady level, the test compounds in the form of free base suspended in 0.5% Methocel (RTM) (90 HG, 15.000, Dow Chem. Corp.), were given intraduodenally at doses from 1 to 8 µmol/kg. The gastric juice was collected by free flow from the gastric cannula in consecutive 30 minutes samples for 3 hours. The samples were titrated to pH 7.0 with 0.1 M NaOH

using a Radiometer automatic titrator and the acid

130 The per cent inhibition of acid secretion was

output was calculated.

calculated by comparing in each dog the acid output in the tests to the acid output in control tests when

only the vehicle was given. The peak inhibitory effect for each compound is given in Table 5 below.

Table 4 Biological effects in isolated rabbit gastric glands

. ∞ .	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	-log IC50
12	so	H	CH ²	Сн3	CH3	CH3	H	Снэ	осн3	CH3	6.5
ŗo	so	H	СНЗ	сн ₃	CH3	н	н	СНЗ	осн3	CH3	6.5
37	SO	н	H	OCH ³	H	н	H	н	_	H ₂) ₄ -	5-0
43	SO	H	н	осн ₂ сн	H	н	H	CH3	OCH3	CH3	4.4
۶l	so	н	H	сн ₂ он	CH3	H	H	CH3	осн3	CH ³	6.1
.04	SO.	Ħ	H	CH2CH2OCH3	H	н	H	CH3	оси3	CH3	5-7
성	\$0	H	сн ₃	осн3	Сн3	н	H	CH3	осн3	CH3	6.5
1	so	Ħ	CH3	OCH ₃	CH3	H	H	CH3	CH3	н	6.7
১ধ	\$O	H	CH3	OCH2CH2OCH3	CH3	н	H	сн3	осн ₃	CH3	5 . 9
60	ŞO	H	CH3	ося2сн2осн3	CH3	н	н	н	CH3	СНЗ	5-4
62	so	H	CH ³	COCH ³	CH3	H	B	СН	осна	CH ₃	6 - 2
64	so	H	сн3	COCH3	CH3	H	H	CH3	н	CH3	5.8
óò	so	н	CH3	COC ₂ H ₅	CH3	H .	H	CH ₃	осн3	сн,	6.0
					_			•	•	•	ont.

cont.

lio .	х	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	-log IC50
68	so	н	CH3	C2HS	CH3	н	н	Сн	осн ₃	сн,	
70	SO	H	CH3	C2H2	CH ₃	н	н		осн	Н	5-9
72	so	H	С ₂ н ₅	CN	C2H5	н	н	CH3	-		5.0
74	SO	Ħ	CH ₃	оснз	CH3	СНЗ	н	CH ³	_	_	6-2
79	so	H	H	CH ₃	H	н	H		осн3	_	
81	so	R	H	-осн ₂ о-		н	н	CH,	осн ₃	CH_	6.1
83	so	н .	-Сн=	сн-сн-сн-	н	н	H	CH3	осн ₃	CH ₃	{5.5 5.3
lú7	50	н	н	осн ₃	н	н	CO2CH3	CH,	0СН,	сн₃ ∫	_
108	so	н.	H	H	осн 3		CO2CH3			CH ₃	3.0

::o.	х	R ¹⁵	RL	R ²	R ³	R ⁴	₹ ⁵	n ⁶	R ¹	R ⁸	-log IC ₅₀
10	so	н•	C#3	сн3	СНЗ	CH ₃	Ħ	CH3	OCH2CH=CH2	CH3	6.1
14	50	н	СНЗ	сн ₃	CH ₃	н	н	СНЗ	OCH2CH=CH2	СНЗ	6.1
ra	so	Н	CH 3	CH3	н	CH 3	н	снз	OCH2CH=CH2	СНЗ	5.9
10	50	н	CH3	cii ³	н	CH3	н	CH3	осн3	СНЗ	6.0
22	so	H	CH3	CH3	H	н	ii	СНЗ	OCH2CH=CH2	CH 3	6.0
24	SQ	H	H	сн ₃	CH3	H	н	CII3	осн ₂ сн=сн ₂	СНЗ	6-0
26	so	H	CH3	н	н	CH3	н	СНЗ	CCH2CH=CH2	СНЗ	5.9
28	so	н	CH3	H	н	н	н	CH3	OCH2CH=CH2	СНЗ	5.9
30	so	н	H	сн ₃	н	н	н	-	CCH2CH-CH2	-	5.9
32	so	н	Н	осн3	н	н	H	CH3	осн,сн=сн,	CH.3	5.6
34	so	н	н	осн ₃	н	н	H	CH3	OCH ₂ C≅CH	CH	5.0
35	so	H	н	осн	н	н	H	R	OCH ₃	c,us	5.6
41	so	н	СНЗ	н	СНЗ	H	н	СНЗ	_	CH	5.9
45	50	н	н -	\bigcirc	H	н	H	_	осн ₃	CH ₃	6.1
									. ***		cont.

cont

жо.	х	R ¹⁵	RL	k ²	R ³	R ⁴	к ⁵	R ⁶	R ⁷	R ⁸	-log IC ₅₀
55	so	h	Ħ	cuoch ₃	CH ₃	н	н	СНЗ	OCH2CH=CH2	СНЗ	5.3
87	so	h	-сн	2 ^{CH} 2 ^{CH} 2 ^{CH} 2	H	н	H	CH3	och ₃	CH3	6.3
91	20	H	h	осн ₂ сн ₂ сн ₂ о-⊙	h	H	Ħ	CH3	осн ₃	CH3	5.8
2	SO	H	CH3	O(CH ₂)6CH3	СН3	H	H	СН3	осн3	СНЗ	5.9
94	so	n	n	^C 2 ^H 5	H	H	H	СНЗ	OCH2CH=CH2	снз	6.6
96	\$0	H	H	oca3	H	h	H	СНЗ	осн ₂ сн ₂ сн(сн ₃) 2	CH3	6.1
98	S0	n	-CH	-Сн-Сн-Ссн ₂ Сн ₂ -		н	H	сн3	осн ₃	СНЗ	5.6
102	SC	н	Н	c(cH3)3	H	н	н	сн3	och ² ch=ch ²	CH ₃	5.9
104	\$0	н	н	CH ⁵ CH ⁵ OCH ³	H	н	н	CH3	осн ₃	СНЗ	5.7
106	so	н	н	-0	`0-	н	н	CH3	OCH3	сн3	. 6.0
111	so	н	н	CH(CH ₃) ₂	н	н	н	снз	OCH ₂	сн3	6.2
113	SO	н	H	сн2сн5сосн3	н	н	н	СНЗ	OCH2CH=CH2	сн3	5.8
118	so	н	н	0 —⟨ ○ ⟩	н	н	Ĥ	čн _З	осн ₃	снз	6.4

No.	X	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	a ⁶	R ⁷	R ⁸	-log IC ₅₀
120	so	Н	н	OCH ₂ CH ₂	Н	н	н	СНЗ	осн ₃	СНЗ	6.3
124	SO.	н	н	- ⊘ .	H	н	н	сн3	осн ₃	снз	7.0
129	so	н	н	Br	н	н	н	снз	осн ₂ сн=сн ₂	сн ₃	
142	SO	н	н	-00	:H ₂ 0-	Н	н	СНЗ	CH3	CH ³	6.0
143	so	н	н	C OCH3	снз	н	н	н	осн ₃	с ₂ н ₅	6.1
145	so	н	сн ₃	сн ₃	сн3	н	н	снз	сн3	н	6.2
147	SO	н	CH3	сн ₃	снз	н	н	н	CH ₃	CH3	6.4
149	SO	н	СНЗ	СНЗ	CH3	н	н	CH3	H	СНЗ	6.2
151	S0	н	СНЗ	сн ₃	н	СНЗ	н	CH3	СНЗ	н	6.3
153	SO	н	CH3	CN	CH3	н	н	CH3	0C2H5	CH3	5.2
77	so	н	н	снз	снз	н	н	н	OCH ₃	C2H5	6.0
159	so	H	H	CF ₃	н	н	н	CH3	$\infty \pi_{\overline{2}} $	сн3	6.3

Table 5 Biological effects in conscious dogs

No.	x	R ¹⁵	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	R ⁷	R ⁸	(I.D.) DOSE	(µmol/kg) % INHIB
84	s	н	н		-CH=CH-CH=CH-	н	н	СН3	осн3	сн3	8	· 85
109	s	н	н	SCH ₃	н	н	н	CH ₃	OCH ₃	CH ₃	8	60

Comment to the test results

It is seen in Table 4 and Table 5 that the tested compounds potently inhibited gastric acid secretion both in vitro and in vivo.

5 CLAIMS

1. A compound of the formula

wherein

10 R^1 , R^2 , R^3 and R^4 , which are the same or different, are

- (a) H
- (b) halogen
- (c) --CN
- (d) --CHO
- 15 (e) --- CF₃

- (g) —O—C—R" (h) —CH(OR¹³)₂
- (i) $-(Z)_0 A C$
- 20 (j) aryl
 - (k) aryloxy
 - (I) alkylthio containing 1-6 carbon atoms
 - (m) -NO₂
 - (n) alkylsulfinyl containing 1-6 carbon atoms or
- 25 wherein

(o) adjacent groups R¹, R², R³ and R⁴ together with the adjacent carbon atoms in the benzimidazole ring form a 5-, 6- or 7-membered monocyclic ring or a 9-, 10- or 11-membered bicyclic ring which rings may be

- 30 saturated or unsaturated and may contain 0-3 hetero atoms selected from —N— and —O—, and which rings may be optionally substituted with 1-4 substituents selected from alkyl groups with 1-3 carbon atoms, alkylene radicals containing 4-5 carbon atoms
- 35 giving spiro compounds, or two or four of these substituents together form one or two oxo groups.

O _.

(—C—), whereby if R¹, R², R³ and R⁴ together with the adjacent carbon atoms in the benzimidazole ring form two rings they may be condensed with each other, in which formulas R¹¹ and R¹², which are the same or different, are

- (a) aryi,
- (b) alkoxy containing 1-4 carbon atoms.
- (c) alkoxyalkoxy containing 1-3 carbon atoms in each alkoxy part,
- (d) arylalkoxy containing 1-2 carbon atoms in the alkoxy part,
 - (e) aryloxy,
 - (f) dialkylamino containing 1-3 carbon atoms in each alkyl residue, or
- (g) pyrrolidino or piperidino, optionally substituted with alkyl containing 1-3 carbon atoms; R¹³ is (a) alkyl containing 1-4 carbon atoms, or
 - (b) alkylene containing 2-3 carbon atoms;

20 nis0or1;

Ais (a) alkylene containing 1-6 carbon atoms

- (b) cycloalkylene containing 3-6 carbon atoms-
- (c) alkenylene containing 2-6 carbon atoms
- (d) cycloalkenylene containing 3-6 carbon atoms,

25 or

(e) alkynylene containing 2-6 carbon atoms; D is (a) —CN

30 wherein

R9 is (a) alkoxy containing 1-5 carbon atoms, or

(b) dialkylamino containing 1-3 carbon atoms in each alkyl residue;

m is 0 or 1;

35 ris0 or 1;

Yis (a) -0-

- (b) —NH—
- (c) -NR10-;

R10 is (a) H

- (b) alkyl containing 1-3 carbon atoms,
 - (c) arylalkyl containing 1-2 carbon atoms in the alkyl part, or
 - (d) aryl;

R⁵ is (a) H or

45 (b) -C-R14;

wherein

R¹⁴ is (a) alkyl containing 1-6 carbon atoms,

- (b) arylalkyl containing 1-2 carbon atoms in the alkyl part
- io (c) aryl
 - (d) alkoxy containing 1-4 carbon atoms
 - (e) arylalkoxy containing 1-2 carbon atoms in the alkyl part
 - (f) aryloxy
- 55 (g) amino

- (h) mono- or dialkylamino containing 1-4 carbon atoms in each alkyl residue
- (i) arylalkylamino containing 1-2 careen atoms in the alkyl part

60 (j) arylamino:

R⁶ and R², which are the same or different, are

(a) Hor

65

- (b) alkyl containing 1-5 carbon atoms; R⁷ is (a) H
- (b) alkyl containing 1-8 carbon atoms
- (c) alkoxy containing 1-8 carbon atoms
- (d) alkenyloxy containing 2-5 carbon atoms
- (e) alkynyloxy containing 2-5 car .n atoms
- (f) alkoxyalkoxy containing 1-2 carbon atoms in each alkoxy group
 - (g) dialkylaminoalkoxy containing 1-2 carbon atoms in each of the alkyl residues on the amino nitrogen and 1-4 carbon atoms in the alkoxy group
- (h) oxacycloalkyl containing one oxygen atom and3-7 carbon atoms
 - (i) oxacycloalkoxy containing two oxygen atoms and 4-7 carbon atoms
 - (j) oxacycloalkylalkyl containing one oxygen atom and 4-7 carbon atoms
- (k) oxacycloalkylalkoxy containing two oxygen atoms and 4-6 carbon atoms, or
- (I) R⁶ and R⁷, or R⁷ and R⁸ together with the adjacent carbon atoms in the pyridine ring from a ring wherein the part constituted by R⁶ and R⁷, or R⁷ and R⁸, is

-CH=CH-CH=CH-

-O-(CH₂)_p-

-CH₂(CH₂)_p--O-CH=CH-

-NH-CH=CH-

-N-CH=CH-

l CH₃

wherein p is 2, 3 or 4 and the O and N atoms always

95 are attached to position 4 in the pyridine ring;
and physiologically acceptable saits of the compounds I wherein X is S;
with the provisos that

- (a) not more than one of R⁵, R⁷ and R⁸ is hydrogen,
- (b) when X is SO, R⁵ is H and R⁶, R⁷ and R³ are selected only from hydrogen, methyl, methoxy, ethoxy, methoxyethoxy and ethoxyethoxy and at the same time more than one of R¹, R², R³ and R⁴ are hydrogen, then those radicals R¹, R², R³ and R⁴ which

105 are not H cannot be selected only from alkyl groups, halogen, alkoxycarbonyl, alkoxy or alkanovi.

- (c) when X is S, R⁵ is H, alkanoyl or alkoxycarbonyl, and R⁶, R⁷ and R⁸ are selected only from hydrogen, methyl, ethyl, methoxy, ethoxy, methoxyethoxy and
- 110 ethoxyethoxy and at the same time more than one of R¹, R², R³ and R⁴ are hydrogen, then those radicals R¹, R², R³ and R⁴ which are not H cannot be selected only from alkyl groups, halogen, alkoxycarbonyl, alkoxy, alkanoyl, trifluoromethyl, or NO₂,
- (d) when X is SO, one of R⁶, R⁷ and R⁸ is H and the other two of R⁶, R⁷ and R⁸ are alkyl, and at the same time more than one of R¹, R², R³ and R⁴ are hydrogen, then those radicals R¹, R², R³ and R⁴ which are not H cannot be selected only from alkyl, halogen, cyano,

(e) when R^3 , R^4 , R^5 and R^{15} are H and simultaneously R^6 and R^8 are H or CH_3 and R^7 is OCH_3 , then R^1 is not 5 CF_3 when R^2 is H, and R^2 is not CF_3 when R^1 is H.

2. A compound according to claim 1 wherein X=S.

3. A compound according to claim 1 wherein X=SO.

4. A compound according to any one of the preceding claims wherein R⁵=H.

5. A compound according to any one of the preceding claims wherein R¹⁵=H.

 A compound according to any one of the
 preceding claims wherein at least three of the radicals R¹, R², R³ and R⁴ are other than hydrogen, or they form at least one ring.

A compound according to any one of the preceding claims wherein R¹, R², R³ and R⁴ are
 selected from H, alkyl and alkoxy groups.

8. A compound according to any one of the preceding claims wherein R^6 and R^8 are selected from H, CH_3 , C_2H_5 , C_3H_7 , $CH(CH_3)_2$ and ring structures connecting with position 4 in the pyridine ring.

9. A compound according to any one of the preceding claims wherein two of the radicals R⁶, R⁷ and R⁸ form one ring structure and the third radical of R⁶, R⁷ and R⁸ is H or alkyl.

A compound according to any one of claims
 1-8 wherein R⁵ and R¹⁵ are H; at reast three of the radicals R¹, R², R³ and R⁴ are other than H; R⁶ and R⁸ are each H or CH₃; and R⁷ is CH₃, OCH₃ or OCH₂CH=CH₂.

11. A compound of the formula:

35 wherein X is S or SO

· R2 is CH3, C2H5, CH(CH3)2 or OCH3.

12. A process for the preparation of a compound of the formula:

$$\begin{array}{c}
R^{2} \\
R^{3}
\end{array}$$

$$\begin{array}{c}
R^{4} \\
R^{5}
\end{array}$$

$$\begin{array}{c}
R^{6} \\
R^{15}
\end{array}$$

$$\begin{array}{c}
R^{6} \\
R^{15}
\end{array}$$

$$\begin{array}{c}
R^{8} \\
R^{15}
\end{array}$$

wherein R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 , R^8 and R^{15} are as 40 defined in claim 1, and X is SO by

oxidizing a compound of the formula I,

wherein R¹⁵, R¹, R², R³, R⁴, R⁵, R⁶, R⁷ and R⁸ have the meanings given above, to give a compound of the same formula I wherein X is S0;

13. Process for preparation of a compound of the formula I wherein R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 , R^8 and R^{15} are as defined in claim 1 and X is S by reacting a compound of the formula:

$$\begin{array}{c|c}
R^2 & R^1 \\
\hline
R^3 & R^4 & R^5
\end{array}$$

50 with a compound of the formula:

$$Z^{2}-CH = N$$

$$Z^{1}$$

$$Z^{1}$$

$$Z^{1}$$

in which formulae R^{15} , R^1 , R^2 , R^3 , R^4 , R^5 , R^6 , R^7 and R^8 are as defined in claim 1 and wherein one of Z^1 and Z^2 is SH and the other is a leaving group, to give a compound of the formula I wherein X is S.

14. Process for the preparation of a compound of the formula I wherein X is S and at least one of R¹, R², R³ and R⁴ is an ester group (Z)_n-A-COOR⁹, COOR¹⁰ or (Z)_n-A-OCOR¹⁰ wherein Z, n, A, R⁹ and R¹⁰ are as defined in claim 1 by esterification of a compound of the formula:

$$R^{8} \xrightarrow{R^{7}} R^{5} \xrightarrow{Y^{1}} Y^{2}$$

$$\downarrow R^{15} \qquad \downarrow N \qquad \downarrow N$$

wherein R¹⁵, R⁵, R⁶, R⁷ and R⁸ are as defined in claim 1 and Y¹, Y², Y³ and Y⁴ represent either R¹, R², R³ and R⁴ as defined in claim 1, respectively, or the groups (Z)_n-A-COOH, COOH and (Z)_n-A-OH, but at least one of Y¹, Y², Y³, Y⁴ is in the acid or alcohol form, by reaction with the appropriate alcohol R⁹OH, R¹⁰OH or carboxylic acid R¹⁰COOH, respectively, to form the required compound.

715. Process for preparation of a compound of the formula I wherein R⁵ is R¹⁴CO and R¹⁴ is as defined in claim 1, by acylation of a compound of the formula:

wherein \overline{R}^{15} , X, R^1 , R^2 , R^3 , R^4 , R^6 , R^7 and R^8 are as defined in claim 1, by reaction with an appropriate acylating agent ($R^{14}CO$) 2O , or $R^{14}CO$ X 1 , wherein X 1 is a leaving group.

 Process for the preparation of a compound of the formula I wherein R⁵ is H, by hydrolyzing a compound of the formula

$$\begin{array}{c|c}
R^{6} & R^{7} \\
R^{6} & R^{1}
\end{array}$$

$$\begin{array}{c|c}
R^{1} & R^{2} \\
R^{15} & R^{3}
\end{array}$$
VI

wherein X, R¹⁵, R¹, R², R³, R⁴, R⁶, R⁷ and R⁸ are as defined in claim 1 and Z³ is a suitable N-protecting 10 group to form the required compound.

- 17. A process according to any one of claims 13-16 wherein a compound in which X is S is obtained and the resulting compound is converted into a physiologically acceptable salt.
- 15 18. A process according to any one of claims 12-17 substantially as hereinbefore described with reference to any one of the Examples.
- A pharmaceutical composition containing a compound or salt according to any of claims 1-11
 together with an inert carrier or diluent.
 - 20. A composition according to claim 19 substantially as hereinbefore described with reference to any one of Examples 167-169.
- A compound according to any one of claims
 1-11 or a physiologically acceptable salt thereof or a composition according to claim 19 or 20 for use in a method of treatment of the human or animal body by surgery or therapy.
- A compound according to any one of claims
 1-11 or a physiologically acceptable salt thereof or a composition according to claim 19 or 20 for use in the treatment of gastric disorders.
- A compound as defined in any of claims 1-11, or a therapeutically acceptable salt thereof, or a
 composition according to claim 19 or 20 for use in inhibiting gastric acid secretion in the human or animal body.
- A compound as defined in any of claims 1-11, or a therapeutically acceptable salt thereof, or a
 composition according to claim 19 or 20 for use as a gastrointestinal cytoprotecting agent in the human or animal body.
- 25. A compound as defined in any of claims 1-11, or a therapeutically acceptable salt thereof, or a
 45 composition according to claim 19 or 20 for use in the treatment of gastrointestinal inflammatory diseases in the human or animal body.
 - 26. A compound of the formula:

$$R^{2a} \xrightarrow{R^{1a}} N \xrightarrow{N} Z^{1a} VIII$$

wherein R^{1a} , R^{2a} , R^{3a} and R^{4a} are the same or different 50 'and selected from the groups

- (a) H,
- (b) alkyl containing 1-6 carbon atoms including cycloalkyl
- (e) alkoxyalkyl containing 1-3 carbon atoms in the
 55 alkoxy residue and 1-6 carbon atoms in the alkyl residue,
 - (d) aryloxyalkyl containing 1-6 carbon atoms in the alkyl residue.
- (e) arylalkyl containing 1-6 carbon atoms in the60 alkyl residue,
 - (f) aryl,
 - (g) alkoxy containing 1-6 carbon atoms,
- (h) alkoxyalkoxy containing 1-3 carbon atoms in the outer alkoxy residue and 1-6 carbon atoms in the
 65 alkoxy residue nearest the aromatic ring.
 - (i) aryloxyalkoxy containing 1-6 carbon atoms in the alkoxy residue,
 - (j) arylalkoxy containing 1-6 carbon atoms in the alkoxy residue, and
- 70 (k) aryloxy, R⁵∎is(a) H,
 - (b) alkoxycarbonyl containing 1-4 carbon atoms in the alkoxy residue,
- (c) arylalkoxycarbonyl containing 1-2 carbon 75 atoms in the alkoxy residue,
 - (d) dialkylaminocarbonyl containing 1-4 carbon atoms in each alkyl residue, or
 - (e) arylaminocarbonyl, and Z^{1a} is (a) SH,
- 60 (b) Cl or Br provided that not more than one of R^{1a}, R^{2a}, R^{3a} and R^{4a} is H.
 - 27. A compound of the formula:

wherein R6a and R8a are

(a) Hor

85

- (b) alkyl containing 1-5 carbon atoms, and R^{7a} is (a) alkenyloxy containing 2-5 carbon atoms, or
 - (b) alkynyloxy containing 2-5 carbon atoms,
- (c) oxacycloalkyl containing one oxygen afom and
 3-7 carbon atoms,
 - (d) oxacycloalkoxy containing two oxygen atoms and 4-7 carbon atoms,
- (e) oxacycloalkylalkyl containing one oxygen atom 95 and 4-7 carbon atoms
 - (f) oxacycloalkylalkoxy containing two oxygen atoms and 4-6 carbon atoms, or
- (g) R^{6a} and R^{7a}, or R^{7a} and R^{8a} together with the adjacent carbon atoms in the pyridine ring form a ring 100 wherein the part constituted by R^{6a} and R^{7a} or R^{7a} and R^{8a} is
 - -CH=CH-CH=CH-
 - -----(CH₂)_{pa}---
 - ---CH₂---(CH₂)_{pa}---
- 105 —O—CH=CH—

wherein pais 2,3 or 4 and the O atom always is attached to position R^{7a},

- and Z^{2a} is (a) SH,
- ·····(b) halogen Cl, Br, I or

(c) OH provided that not more than one of R^{6a} and R^{8a} is H.

Printed for Her Majesty's Stationery Office by The Tweeddale Press Ltd., Berwick-upon-Tweed, 1984. Published at the Patent Office, 25 Southampton Buildings, London WC2A 1AY, from which copies may be obtained.